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Pressure-Driven Metallization in Hafnium Diselenide

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5 **ABSTRACT:** The quest for new transition metal dichalcogenides (TMDs) with 6 outstanding electronic properties operating under ambient conditions draws us to 7 investigate the 1T-HfSe₂ polytype under hydrostatic pressure. Diamond anvil cell 8 (DAC) devices coupled to *in situ* synchrotron X-ray, Raman, and optical (vis– 9 NIR) absorption experiments along with density functional theory (DFT)-based 10 calculations prove that (i) bulk 1T-HfSe₂ exhibits strong structural and vibrational 11 anisotropies, being the interlayer direction especially sensitive to pressure changes, 12 (ii) the indirect gap of 1T-HfSe₂ trend to vanish by a -0.1 eV/GPa pressure rate, 13 slightly faster than MoS₂ or WS₂, (iii) the onset of the metallic behavior appears at 14 $P_{met} \sim 10$ GPa, which is to date the lowest pressure among common TMDs, and



15 finally, (iv) the electronic transition is explained by the bulk modulus B_0 - P_{met} correlation, along with the pressure coefficient of the 16 band gap, in terms of the electronic overlap between chalcogenide p-type and metal d-type orbitals. Overall, our findings identify 1T-17 HfSe₂ as a new efficient TMD material with potential multipurpose technological applications.

18 INTRODUCTION

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19 Lavered-based materials have attracted tremendous research 20 interest since the breakthrough of graphene.¹ These materials 21 have been widely studied owing to the intrinsic electronic 22 properties covering from semiconductivity to metallic behavior 23 along with promising catalytic performance, photolumines-24 cence, and high-water resistance.²⁻⁵ Particularly, layered 25 transition metal dichalcogenides (TMDs) exhibit unique 26 optical and electronic properties accompanied by a tunable 27 band gap.⁶ From the structural point of view, TMDs are 28 defined as solids with the general stoichiometry MX₂ (M is a 29 transition metal that belongs to the groups IV, V, VI, and VII; $_{30}$ X = S, Se, Te) showing in-plane strong chemical bonding and 31 out-of-plane weak van der Waals (vdW) interactions, as 32 represented in Figure 1a. Based on the atomic arrangements, 33 TMDs are classified in three stacking polytypes denoted as 1T, 34 2H, and 3R that correspond to one, two, and three layers per 35 unit cell with a trigonal, hexagonal, and rhombohedral 36 structure, respectively. Note, however, that M₂X₃ and MX 37 alternative stoichiometries have been recently found for 38 TMDs.^{7,8} There are roughly 60 TMDs known so far, but 39 since MoS_2 , TiS_2 , and WS_2 have attracted more attention, 40 these three compounds are considered the workhorses among 41 them.^{9,10}

⁴² Not surprisingly, the structural and electronic properties of ⁴³ few layers and bulk TMDs can be easily tuned thanks to the ⁴⁴ abovementioned vdW interlayer forces.¹¹ Several strategies ⁴⁵ have been established to enhance their performance including ⁴⁶ (i) top-down synthesis protocols moving toward low-dimen-⁴⁷ sional TMD counterparts,¹² (ii) the use of chemical dopant agents,¹³ (iii) the intercalation of species,¹⁴ (iv) the application $_{48}$ of an electrical field,¹⁵ and (v) the application of high $_{49}$ pressures.¹⁶ The latter strategy brings the layers closer and thus $_{50}$ making the interlayer interactions much stronger, resulting in a $_{51}$ concomitant modulation of the TMD properties. $_{52}$

Recent experimental and theoretical investigations on the 53 pressure-induced properties of bulk TMDs, such as MoS2, 54 WS₂, MoSe₂, WSe₂, and MoTe₂, have reported interesting 55 semiconducting-to-metal electronic state transitions achieved 56 at \sim 15–40 GPa under hydrostatic conditions, depending on $_{57}$ the TMD composition.¹⁷⁻²³ The implementation of these ₅₈ results in flexible nanoelectronic devices involves practical 59 difficulties mainly due to the high-pressure regimes required to 60 induce the metallization.²⁴ To overcome this drawback, low- 61 dimensional counterparts of TMDs coupled with uni- and 62 biaxial stress conditions have been proposed as a suitable 63 alternative to modulate the band gap at lower pressures. 64 Combined experimental and theoretical studies have success- 65 fully shown that the metallization of monolayer MoS₂ takes 66 place only at 3 GPa under uniaxial out-of-plane compression 67 conditions.^{25,26} 68

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Figure 1. View of the crystal structure of 1T-HfSe₂ showing (a) the multilayered arrangement and (b) the *a-b* plane. (c) Schematic of the used high-pressure diamond anvil cell setup. (d) Observed (circles), calculated (solid line), and difference (bottom) synchrotron-XRD Rietveld profile for 1T-HfSe₂ at room temperature and ambient pressure. Bragg positions indicated in green correspond to 1T-HfSe₂, whereas orange marks are due to a very small amount of impurity (less than 3%) identified as metallic Se.

All these extensive efforts have contributed significantly to 69 the knowledge of bulk, few layer, and monolayer TMDs under 70 extreme conditions. Most of the studies have been focused 71 nly on the 2H-TMD polytype, thus demanding further 72 investigations over TMDs with other arrangements (e.g., 1T or 73 3R). In particular, 1T-TMD compounds have attracted interest 74 for understanding the mechanism of charge density wave order 75 at low temperature and its coexistence with superconductivity 76 under high pressure.²⁷⁻²⁹ Focusing on hafnium diselenide 77 (1T-HfSe₂), depicted in Figure 1a,b, its high carrier mobility 78 (above 2000 cm^2 V s⁻¹), significantly larger than that of 2H-79 $(Mo,W)X_2$ (~340 cm² Vs⁻¹), is to be highlighted. Due to this 80 extraordinary electronic property, 1T-HfSe2 emerges as one of 81 the most promising materials for applications in the ambit of 82 ⁸³ field-effect transistors (FETs).^{30,31} Furthermore, the moderate $_{84}$ band gap of 1T-HfSe₂ (~1.1 eV)³² opens the possibility of $_{85}$ using it as a high- κ dielectric, leading to the replacement of 86 silicon in electronic devices.³³

In short, the present study has the aim of providing 87 88 experimental data and theoretical interpretation on the response of structural, vibrational, and electronic properties 89 of bulk 1T-HfSe₂ at hydrostatic high pressures. A homemade 90 diamond anvil cell (DAC) device has been used for performing 91 in situ high-pressure measurements, and therefore, synchrotron 92 X-ray diffraction, Raman spectroscopy, and optical (vis-NIR) 93 absorption experiments have been carried out. In addition, 94 density functional theory (DFT)-based calculations assist and 95 96 interpret the experiments. Our investigations demonstrate that 97 the 1T-HfSe₂ phase is highly sensitive under pressure, 98 reporting a metallization at ~10 GPa, which is actually the 99 lowest value reported to date for bulk TMDs under hydrostatic 100 conditions. Note that this relatively low hydrostatic pressure 101 may break down the technological limitations observed in $_{102}$ other compounds within the 2H-(Mo,W)(S, Se)₂ polytype 103 family, opening new avenues in the generation of novel flexible 104 nanoelectronic devices.

EXPERIMENTAL AND COMPUTATIONAL DETAILS 105

Samples. $HfSe_2$ samples used in the experiments were acquired 106 from HQGraphene in a single crystal form. A pristine single crystal 107 was mechanically exfoliated and cut into smaller pieces to obtain 108 various samples of similar dimensions and suitable to be loaded in the 109 high-pressure cells. The crystal structure was checked by room- 110 temperature XRD, and purity and homogeneity were probed by 111 scanning electron microscopy (SEM) and X-ray energy-dispersive 112 spectroscopy (XEDS). Further details are given in Figure S1 in the 113 Supporting Information. 114

High-Pressure Experiments. Symmetric diamond anvil cells 115 (DAC) were used for high-pressure experiments. The general 116 operating scheme of the DAC is depicted in Figure 1c. In situ high- 117 pressure synchrotron X-ray diffraction experiments were performed at 118 beamline ID15B (experiment CH-5079) of the European Synchro- 119 tron Radiation Facility. A small piece of HfSe₂ was ground into fine 120 powder, subsequently loaded in a membrane-type DAC with 300 μ m 121 culet-sized diamonds. A stainless steel gasket was preindented to be 122 50 μ m thick, and a hole of 100 μ m in diameter was drilled. A sample 123 chamber was filled with helium as a pressure transmitting medium 124 (PTM) and two ruby chips for calibrating pressure. The beam was 125 operated at 0.411 Å with dimensions close to 5 \times 5 μ m². Angle- 126 dispersive X-ray patterns were collected with a MAR555 image plate 127 detector. Images were integrated by using both Fit2D and Dioptas 128 software. Rietveld analysis of data was performed through Fullprof 129 software.34,35 130

High-pressure Raman experiments were carried out using a 131 Merril–Basset DAC device with a 400 μ m culet size. In this case, a 132 small single crystal flake of HfSe₂ was loaded into the chamber sample 133 of a stainless-steel gasket as well as Daphne 7373 oil as the PTM and 134 ruby chips as pressure markers. An XploRA PLUS confocal Raman 135 spectrometer equipped with excitation lines at 532 and 785 nm and 136 several long-distance working objectives was used to acquire the 137 Raman spectra.

Optical absorption under high-pressure conditions was performed 139 on a prototype fiber optics microscope equipped with two 25×140 reflecting objectives mounted on two independent *xyz* translation 141 stages for a microfocus beam, the collector objective, and a third 142 independent *xyz* translation stage for DAC micropositioning. Optical 143 absorption data and images were obtained simultaneously with the 144 same device. Spectra in the UV–vis and NIR were recorded with an 145 Ocean Optics USB 2000 and NIRQUEST 512 monochromators 146 using Si- and InGaAs-CCD detectors, respectively. To ensure a good 147

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Figure 2. Experimental (from XRD) and theoretical pressure evolution of (a) normalized a/a_0 and c/c_0 lattice parameters and (b) volume of the 1T-HfSe₂ unit cell; third-order BM-EOS is also shown in panel (b). (c) Ambient pressure Raman spectrum of 1T-HfSe₂, showing the two main vibrational modes with E_g and A_{1g} symmetry. (d) Evolution of Raman shift frequencies with pressure.

148 quality signal, the absorption spectra were obtained with an $HfSe_2$ 149 single crystal flake with a thickness of around 60 μ m. Analysis of data 150 and band gap calculation methodology is detailed in Figure S2 and 151 Note S1.

Density Functional Theory-Based Calculations. First-princi-152 153 ples periodic electronic structure calculations were systematically 154 performed by minimizing static total energies at selected volumes of 155 the 1T-HfSe2 structure. To this end, the Vienna ab initio simulation 156 package (VASP)³⁶ was used by employing the Perdew-Burke-157 Ernzerhof (PBE) implementation³⁷ within the generalized gradient 158 approximation (GGA) to the exchange-correlation (xc) functional. 159 Weak vdW interactions were considered by adding a semiempirical 160 dispersion potential to the conventional Kohn-Sham DFT energy 161 through a pairwise force field following Grimme's DFT-D3 162 method.^{38,39} This dispersion term is highly required to obtain a 163 reasonable agreement with respect to experiments even in dense 164 solids.⁴⁰ The Kohn-Sham equations were solved by using an 165 expansion of the valence electron density in a plane-wave basis set 166 with a kinetic energy cutoff of 500 eV. The projector-augmented wave 167 (PAW) method was included to account for the interaction between 168 the valence and the core electron densities.^{41,42} Numerical 169 integrations in a reciprocal space were carried out by sampling Γ -170 centered Monkhorst-Pack meshes,43 where the numbers of 171 subdivisions along each reciprocal lattice vector \mathbf{b}_i were given by N_i $\max(1.90 |\mathbf{b}_i| + 0.5)$. The geometry optimizations were considered 172 = converged when the forces acting on the nuclei were all below 10^{-5} 173 174 eV Å⁻¹. Note, however, that this criterion is even tight, 10^{-8} eV Å⁻¹, 175 to calculate the phonon frequencies of Raman active modes at the Γ -176 point by using finite differences, as implemented in the VASP 177 package.

In addition, since the standard GGA calculations systematically 179 underestimate the band gap in semiconductor materials, a more 180 sophisticated approach as one based on a hybrid functional would be 181 recommended.⁴⁴ A state-of-the-art hybrid-based calculation strategy 182 was followed in this regard to accurately investigate the electronic 183 properties of 1T-HfSe₂. Thus, the hybrid HSE06 density functional 184 was selected.⁴⁵ By using the hybrid HSE06 functional, 25% of the 185 short-range exchange interaction of the traditional PBE *xc* functional 186 was replaced by the short-range nonlocal Hartree–Fock exchange interaction. In addition, an exchange-screening parameter ω of 0.2 187 Å⁻¹ was applied. Particular numerical integrations by sampling Γ - 188 centered Monkhorst–Pack meshes with $N_i = \max(1.30 \text{ lb}_i^{-1} + 0.5)$ 189 were performed.

RESULTS AND DISCUSSION

Structural Characterization. To investigate the structure 192 of 1T-HfSe2 and its evolution under hydrostatic pressure 193 conditions, in situ high-pressure synchrotron X-ray diffraction 194 (XRD) and Raman spectroscopy analysis coupled with DFT- 195 based calculations are carried out. The representative XRD 196 pattern for 1T-HfSe2 at ambient pressure depicted in Figure 1d 197 shows a high crystallinity of the sample. The Rietveld 198 refinement confirms that 1T-HfSe₂ belongs to the P3m1 199 space group containing just one HfSe2 formula unit per unit 200 cell with trigonal symmetry (see Table S1 for details). The cell 201 parameters are a = 3.7562 Å and c = 6.1749 Å. The internal 202 coordinates are (0,0,0) and (1/3,2/3,0.2560) for Hf and Se, 203 respectively. DFT-based calculations predict a = 3.7319 Å, c = 2046.2152 Å, and $z_{Se} = 0.2538$, which are in good agreement with 205 the experiments. This confirms that the PBE-D3 level of 206 calculation is reasonably reliable to describe the structure of 207 1T-HfSe₂. 208

As applied pressure increases, the XRD patterns show the 209 well-known shifting toward higher 2θ angles (see Figure S3). 210 New peaks are not observed during the compression process 211 up to 11.3 GPa, and the initial 1T-HfSe₂ structure is fully 212 recovered at ambient pressure. Consequently, the compression/decompression cycle is completely reversible (see Figure 214 S3 and Note S2). This outstanding mechanical stability is one 215 of the main desired requirements for the development of future 216 electronic devices operating under harsh conditions. 217

At this respect, we would like to notice that a pressure- 218 induced phase transition to a C2/m monoclinic structure was 219 recently reported in 1T-TiTe₂ showing coexistence with the 220 trigonal phase in a wide range of pressures from ~5 to 19 221



Figure 3. (a) Pressure evolution of the vis–NIR transmittance spectra for 1T-HfSe₂. (b) Optical absorption coefficient versus photon energy at ambient pressure (the inset shows the linear extrapolation to estimate the band gap). (c) Evolution of the experimental and theoretical calculations of the band gap energy with pressure.

222 GPa.⁴⁶⁻⁴⁸ The claimed generality of this transformation in 223 TMD MX₂ compounds was later examined by Mora-Fonz et 224 al.⁴⁹ The emergence of the monoclinic C2/m structure is 225 explicitly ruled out in their exhaustive computational study of 226 the 1T-TiSe₂ polytype. They found that this trigonal structure 227 is the stable phase up to 25 GPa and conclude that the 228 pressure-induced polymorphic sequence is still an open 229 question. For 1T-TiS₂, instead of the C2/m monoclinic 230 structure, a cotunnite-like polymorph is proposed as the high-231 pressure structure where the trigonal polytype transits at 16.2 232 GPa.⁵⁰ The lower oxidation power of Te and the distinctive 233 role of the 5p orbital in Te-based TMDs, compared to S and 234 Se counterparts, should be taken into account to explain the 235 rich polymorphism of the TMD family according to the high-236 pressure study of Léger et al. in 1T-IrTe2.⁵¹ This is of relevance 237 to our work showing that, in all these experimental and 238 theoretical studies, the trigonal 1T polytype is always observed 239 in the pressure range where our experiments and calculations 240 have been performed.

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The experimental and computed pressure dependence of the 241 unit cell volume V and lattice parameters a and c are shown in 242 Figure 2a and Figure 2b, respectively. V, a, and c exhibit 243 244 smooth variations as pressure increases for both experiments and theory. Computational results are in reasonable agreement 245 with the experiments, although at high pressure, the calculated 246 lattice parameter reduction along the interlayer *c*-axis direction 247 is predicted to be slightly higher than the experimental one. 248 This deviation from the experiments can be attributed to the 249 description of the vdW interactions that is reasonable in the 250 251 absence of pressure but is progressively underestimated as pressure increases according to the D3 correction. Overall, the 252 253 calculated trends observed under pressure are well-described 254 compared to experiments.

Quantitatively speaking, the pressure evolution of a, c, and V256 is carried out by fitting analytical functions to the data. The experimental third-order Birch–Murnaghan (BM3) equation ${}_{257}$ of state (EOS) 52,53 (see Figure 2b) provides a zero-pressure ${}_{258}$ volume (V_0) equal to 75.42 Å³ and an isothermal zero pressure ${}_{259}$ bulk modulus (B_0) along with its first pressure derivative (B_0') ${}_{260}$ of 34.5 GPa and 6.9, respectively. DFT calculations lead to V_0 261 = 74.96 Å³, B_0 = 29.2 GPa, and B_0' = 7.0, in reasonable 262 agreement with the experimental results. Here, it must be 263 noted that the low B_0 value compared with other compounds 264 within the MX₂ crystal family (for example, 57 and 62 GPa for 265 MoS₂ and MoSe₂, respectively)^{17,19} shows an interesting result 266 that might anticipate a low metallization pressure for 1T-HfSe₂, 267 assuming that the metallization is a consequence of short- 268 distance interlayer interactions.

Concerning the effect of pressure on the unit cell 270 parameters, it is more pronounced in the c axis than in the a 271 axis. The former is reduced around 8.3%, whereas the latter is 272 only reduced around half of it, ~4.5%, at the maximum 273 reached pressure in our experiments, close to 11 GPa. Linear 274 isothermal compressibility (k_i) values of $k_a = 5.51 \text{ TPa}^{-1}$ and k_c 275 = 21.0 TPa⁻¹ and k_a = 5.51 TPa⁻¹ and k_c = 29.6 TPa⁻¹ are 276 obtained from the experimental and calculated data, 277 respectively. These values are fully consistent with the trends 278 displayed in Figure 2a and evidence a clear pressure-induced 279 structural anisotropy in 1T-HfSe2. Likewise, the pressure 280 evolution of Se-Se and Hf-Se interatomic distances 281 corroborates this observation since the reduction of the Se- 282 Se interlayer distance is much more pronounced than 283 corresponding to the intralayer Hf-Se one (see Figure S4). 284 All these structural results have obvious implications in the 285 electronic behavior of this polytype as we will discuss later. 286

Following with the structural characterization, Raman ²⁸⁷ spectroscopy has been broadly used to characterize TMD- ²⁸⁸ derived materials. ^{54,55} The Raman spectrum of 1T-HfSe₂ ²⁸⁹ (Figure 2c) clearly shows two signals centered at 147 and ²⁹⁰ 199 cm⁻¹ corresponding to normal modes of E_g and A_{1g} ²⁹¹

292 symmetry in good agreement with previous works, respec-²⁹² symmetry in good egreened to in-plane vibrations, ²⁹³ tively.^{56,57} The E_g mode corresponds to in-plane vibrations, 294 whereas the A1g one is associated with out-of-plane atomic 295 movements. Interestingly, the DFT-based calculations give 296 frequencies at 148 and 202 cm⁻¹, respectively. Note that the 297 computed Raman frequency modes match quite well in the 298 absence of pressure. Not surprisingly, both Raman modes shift 299 toward high frequencies at increasing pressure (Figure 2d). $_{300}$ The experimental pressure coefficients for E_g and A_{1g} modes $_{301}$ are 1.67 and 3.33 $\rm cm^{-1}$ GPa^{-1}, respectively. Again, our 302 calculations reproduce reasonably well these values. The weak 303 interactions along the *c*-axis direction are significantly sensitive 304 to pressure effects. This structural argument directly explains 305 why the A_{1g} pressure coefficient is higher compared to that of $_{306}$ the E_g mode. The different response of the vibrational modes, 307 depending on whether they are in-plane or out-of-plane, is another manifestation of pressure-induced anisotropy exhibited 308 309 by this material (see Figure S5 for a detailed scheme of normal 310 modes and the Raman spectra under pressure).

Electronic Characterization. We show now results from 311 312 the analysis of the electronic properties of 1T-HfSe₂ and, more 313 interestingly, their evolution under pressure. To this end, in 314 situ high-pressure optical absorption experiments coupled with 315 DFT-based electron band calculations are carried out. The vis-NIR absorption spectra are analyzed in detail (Figure 3a). 316 First, we focus on the spectrum in the absence of pressure. 317 318 Plotting the optical absorption coefficient square root versus 319 photon energy allows one to estimate the band gap, E_{g} , using a 320 linear extrapolation (see details in Figure S2 and Note S1). In 321 the absence of pressure, 1T-HfSe₂ yields to an indirect gap of 322 1.05 eV estimated by the linear extrapolation (Figure 3b, 323 inset). Our experiments at zero pressure are in good agreement 324 with previous studies.^{33,58,59} In addition, accurate DFT-based 325 calculations using the hybrid HSE06 xc functional reveal two 326 close indirect gaps at Γ -L (0.97 eV) and Γ -M (1.14 eV), as 327 shown in the band structure depicted in Figure 4a. This leads 328 to a theoretical E_{σ} value of 0.97 eV, in close agreement with the 329 experimental one.

The effect of pressure is analyzed based on the vis-NIR 330 331 spectra depicted in Figure 3a. A significant reduction of the 332 transmittance within the transmission window below 1.2 eV is observed as pressure increases. Interestingly, the transmission 333 334 edge becomes almost negligible at 10.2 GPa. This result 335 confirms the closure of the band gap and evidences the 336 metallization of 1T-HfSe2. The linear extrapolation method (see the inset in Figure 3b) has been also used to evaluate the 337 pressure evolution of the band gap. Pressure evolution of the 338 339 band gap energy is shown in Figure 3c. The E_{σ} narrowing 340 progresses steadily and smoothly as the pressure increases. A 341 reduction of 50% in the band gap is already achieved at 4 GPa. The E_{α} pressure evolution exhibits an almost linear trend, easily 342 343 represented with a second-order polynomial, similar to other well-known TMDs such as MoS₂ and MoSe₂.^{17,19} 344

Further interesting results are found in the analysis of the at6 abovementioned calculated indirect gaps at Γ -L and Γ -M regions of the Brillouin zone. The former shows the smallest band gap along the investigated pressure range and disappears at8 at 8.1 GPa, whereas the latter is zero at around 10.9 GPa by an so extrapolation analysis (Figure 3c). These computational results pinpoint the emergence of a metal state observed in the ss2 experiments and confirm that bulk 1T-HfSe₂ becomes metallic ss3 at ~10 GPa, the lowest metallization pressure known so far in ss4 MX₂ TMD materials (see Figure 4b).



Figure 4. DFT-based calculations of the band structures on 1T-HfSe₂ at (a) ambient pressure and (b) metallization pressure. Red lines indicate the Fermi level, and arrows show the indirect band gaps.

To identify the atomic orbitals involved in the metallization 355 of 1T-HfSe₂, the partial density of states (DOS) depicted in 356 Figure 5 is discussed. This analysis reveals a predominant 357 f5 contribution of the Se 4p orbitals in the valence band (VB), 358 whereas the Hf 5d orbitals are the ones dominating the 359 conduction band (CB). When 1T-HfSe₂ is subjected to high 360 pressure, the contribution from the Se p_z in the VB and those 361 of the metal d_z^2 , d_{xz} , and d_{yz} orbitals in the CB become more 362 important, therefore decreasing the participation of the p_x and 363 p_y orbitals of the chalcogen and d_{xy} and $d_{x^2 - y^2}$ of Hf.

This means that the orbitals with electron density localized 365 along the *z* direction increase their contribution in the density 366 of states of the VB and CB by the effect of pressure. This trend 367 is directly connected with the structural anisotropy of this 368 material, as described in the previous structural character- 369 ization. From the chemical point of view, the hydrostatic 370 pressure induces an effective overlapping between the orbitals 371 directed along the *z* direction, and then, noncovalent 372 interactions are strengthened.²⁶ As a consequence, we 373 conclude that the CB is stabilized thanks to the effective 374 overlapping of these orbitals, producing the vanishing of the 375 band gap and the eventual metallization of 1T-HfSe₂ under 376 pressure.

All these results point out that the metallization of 1T-HfSe_2 378 is conditioned mainly by its response to hydrostatic pressure 379 and particularly by the compressibility along the *c* axis. An 380 interesting comparison is schemed in Figure 6a, where the B_0 381 f6 parameter of different TMDs is included along with our result 382 corresponding to 1T-HfSe_2 . The latter presents the lowest B_0 383 values among the selected TMD family. It is probably due to a 384 good compromise between the efficient overlapping of the 385 chalcogen's p orbitals, which are much more effective in 386 selenides than in sulfides, and the large Hf ionic radius in 387 comparison with other metals as Mo or W. Both factors favor a 388 strengthening of noncovalent interactions along the *z* 389 direction, which justifies a higher compressibility and less 390 rigidity of this material. In fact, such a behavior has previously 391

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Figure 5. Partial density of states (DOS) for Hf (left) and Se (right) atoms at 0.3 GPa (upper) and at metallization pressure (lower).



Figure 6. (a) Metallization pressure versus bulk modulus for different TMDs. (b) Evolution of the band gap energy with pressure for different TMDs, showing the metallization pressure for each material.

 392 been theoretically predicted by Guzmán and Strachan for 393 TMDs. 60

These results reveal a correlation between the bulk modulus 394 and the metallization pressure for different TMDs explored in 395 Figure 6a. It is clearly seen that the more compressible is a 396 material, the lower its metallization pressure is expected, and 397 thus, 1T-HfSe₂ has been positioned in a very attractive 398 pressure regime, close to 10 GPa. 2H-MoTe2, 2H-MoS2, 2H- 399 MoW₂, 2H-MoSe₂, and 2H-WSe₂ follow, in this order, 400 increasing values of both B_0 and P_{met} . Analogously, the band 401 gap versus pressure plotted in Figure 6b show that the linear 402 trends (guide to the eye) of $2H-MX_2$ (M = Mo, W; X = S, Se, 403 Te)^{17-20,22,23} and $1T-HfSe_2$ (this work) also follow the same 404 sequence. Clearly, the reduction of E_{σ} in 1T-HfSe₂ occurs at 405 the highest pressure rate, close to 0.1 eV·GPa⁻¹. We thus 406 conclude that 1T-HfSe₂ presents attractive tunable electronic 407 properties that can be easily modulated by the application of 408 relatively moderate pressures. 409

We have undertaken a complete and detailed study on bulk 411 1T-HfSe₂, covering the analysis of its structural, dynamical, and 412 electronic properties up to 12 GPa. We have combined 413 experiments and theoretical calculations to shed light on this 414 TMD material to understand its behavior under pressure. Our 415 results conclude that 1T-HfSe₂ shows a highly sensitive 416 response to pressure application, giving increase to a 417 pressure-driven metallization at around 10 GPa, which is 418 actually the lowest value found to date among common bulk 419 TMD-derived materials. The theoretical calculations are in 420 good agreement with the experimental observations, and the 421 correlation between electronic behavior of the solid and the 422 mechanical response of its crystallographic structure under 423 pressure has been evidenced. Within the studied pressure 424 range, there is no first-order structural transition found, and 425 both structure and electronic behavior are completely 426 reversible, which is an essential requirement for the develop- 427 ment of future electronic devices operating under harsh 428 conditions. This scenario situates 1T-HfSe₂ in a privileged 429

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Author Contributions

495 A.A.-C. and J.S.-B. conceived the idea and led the project. A.A.- 496 C. designed the experiments and performed the synchrotron 497 XRD, Raman, and optical absorption measurements. A.M.-G., 498 M.A.S., P.P., R.F., and J.M.R. conducted the theoretical 499 calculations. G.G. contributed to the synchrotron XRD 500 measurements at high pressure. J.A.B.-A. and J.G. contributed 501 to the optical absorption experiments at high pressure and 502 analysis of data. M.T. and V.G.B. contributed to the analysis, 503 interpretation, and discussion of results. A.A.-C., A.M.-G., 504 J.M.R., and J.S.-B. wrote the manuscript with the help of all the 505 authors. All the authors commented on the final manuscript. 506 J.S.-B. supervised the project. 507

Notes

The authors declare no competing financial interest.

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REFERENCES

(1) Geim, A. K.; Novoselov, K. S. The Rise of Graphene. Nat. Mater. 520 2007, 6, 183-191. 521

(2) Butler, S. Z.; Hollen, S. M.; Cao, L.; Cui, Y.; Gupta, J. A.; 522 Gutiérrez, H. R.; Heinz, T. F.; Hong, S. S.; Huang, J.; Ismach, A. F.; 523 Johnson-Halperin, E.; Kuno, M.; Plashnitsa, V. V.; Robinson, R. D.; 524 Ruoff, R. S.; Salahuddin, S.; Shan, J.; Shi, L.; Spencer, M. G.; 525 Terrones, M.; Windl, W.; Goldberger, J. E. Progress, Challenges, and 526 Opportunities in Two-Dimensional Materials Beyond Graphene. ACS 527 Nano 2013, 7, 2898-2926. 528

(3) Liu, Y.; Weiss, N. O.; Duan, X.; Cheng, H.-C.; Huang, Y.; Duan, 529 X. Van der Waals Heterostructures and Devices. Nat. Rev. Mater. 530 2016, 1, 16042. 531

(4) Centi, G.; Perathoner, S. Catalysis by Layered Materials: A 532 Review. Microporous Mesoporous Mater. 2008, 107, 3-15. 533

(5) Purdie, D. G.; Pugno, N. M.; Taniguchi, T.; Watanabe, K.; 534 Ferrari, A. C.; Lombardo, A. Cleaning Interfaces in Layered Materials 535 Heterostructures. Nat. Commun. 2018, 9, 5387. 536

(6) Choi, W.; Choudhary, N.; Han, G. H.; Park, J.; Akinwande, D.; 537 Lee, Y. H. Recent Development of Two-Dimensional Transition 538 Metal Dichalcogenides and Their Applications. Mater. Today 2017, 539 20. 116-130. 540

(7) Zhang, J.; Peng, Z.; Soni, A.; Zhao, Y.; Xiong, Y.; Peng, B.; 541 Wang, J.; Dresselhaus, M. S.; Xiong, Q. Raman Spectroscopy of Few- 542 Quintuple Layer Topological Insulator Bi2Se3 Nanoplatelets. Nano 543 Lett. 2011, 11, 2407-2414. 544

(8) Zhou, X.; Cheng, J.; Zhou, Y.; Cao, T.; Hong, H.; Liao, Z.; Wu, 545 S.; Peng, H.; Liu, K.; Yu, D. Strong Second-Harmonic Generation in 546 Atomic Layered GaSe. J. Am. Chem. Soc. 2015, 137, 7994-7997. 547 (9) Manzeli, S.; Ovchinnikov, D.; Pasquier, D.; Yazvey, O. V.; Kis, A. 548 2D Transition Metal Dichalcogenides. Nat. Rev. Mater. 2017, 2, 549 17033. 550

430 position compared to other TMDs. In fact, most usual 431 strategies for effectively tuning the electronic properties of 432 these systems concern the dimensionality reduction, working 433 at the nanoscale (monolayers). Through the application of 434 hydrostatic pressure, some materials require even more than 60 435 GPa for inducing a semiconductor-metal electronic transition. 436 In contrast, 1T-HfSe2 would allow us to work within the 437 macroscopic scale, being able to significantly modulate their 438 optoelectronic properties by means of moderate pressure 439 application.

440 ASSOCIATED CONTENT

441 Supporting Information

442 The Supporting Information is available free of charge at 443 https://pubs.acs.org/doi/10.1021/acs.inorgchem.0c03223.

Additional supporting figures, tables, and notes showing 444

- SEM, XEDS, optical density explorations, XRD patterns, 445 pressure evolution, crystallographic data, and other 446
- experiments (PDF) 447

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(10) Chhowalla, M.; Shin, H. S.; Eda, G.; Li, L.-J.; Loh, K.-P.; Zhang,
H. The Chemistry of Two-Dimensional Layered Transition Metal
Dichalcogenide Nanosheets. *Nat. Chem.* 2013, *5*, 263–275.

554 (11) Qiu, X.; Ji, W. Illuminating Interlayer Interactions. *Nat. Mater.* 555 **2018**, *17*, 211–213.

556 (12) Lv, R.; Robinson, J. A.; Schaak, R. E.; Sun, D.; Sun, Y.; Mallouk,

557 T. E.; Terrones, M. Transition Metal Dichalcogenides and Beyond: 558 Synthesis, Properties, and Applications of Single- and Few-Layer 559 Nanosheets. *Acc. Chem. Res.* **2015**, *48*, 56–64.

560 (13) Balasubramaniam, B.; Singh, N.; Kar, P.; Tyagi, A.; Prakash, J.;

561 Gupta, R. K. Engineering of Transition Metal Dichalcogenide-Based
562 2D Nanomaterials Through Doping for Environmental Applications.
563 Mol. Syst. Des. Eng. 2019, 4, 804–827.

564 (14) Jung, Y.; Zhou, Y.; Cha, J. J. Intercalation in Two-Dimensional 565 Transition Metal Chalcogenides. *Inorg. Chem. Front.* **2016**, *3*, 452– 566 463.

(15) Liu, F.; Zhou, J.; Zhu, C.; Liu, Z. Electric Field Effect in Two568 Dimensional Transition Metal Dichalcogenides. *Adv. Funct. Mater.*569 2017, 27, 1602404.

570 (16) Mao, H.-K.; Chen, B.; Chen, J.; Li, K.; Lin, J.-F.; Yang, W.; 571 Zheng, H. Recent Advances in High-Pressure Science and 572 Technology. *Matter Radiat. Extremes* **2016**, *1*, 59–75.

573 (17) Nayak, A. P.; Bhattacharyya, S.; Zhu, J.; Liu, J.; Wu, X.; Pandey,

574 T.; Jin, C.; Singh, A. K.; Akinwande, D.; Lin, J.-F. Pressure-Induced 575 Semiconducting to Metallic Transition in Multilayered Molybdenum 576 Disulphide. *Nat. Commun.* **2014**, *5*, 3731.

577 (18) Nayak, A. P.; Yuan, Z.; Cao, B.; Liu, J.; Wu, J.; Moran, S. T.; Li, 578 T.; Akinwande, D.; Jin, C.; Lin, J.-F. Pressure-Modulated Con-579 ductivity, Carrier Density, and Mobility of Multilayered Tungsten 580 Disulfide. *ACS Nano* **2015**, *9*, 9117–9123.

(19) Zhao, Z.; Zhang, H.; Yuan, H.; Wang, S.; Lin, Y.; Zeng, Q.; Xu,
Sc, Liu, Z.; Solanki, G. K.; Patel, K. D.; Cui, Y.; Hwang, H. Y.; Mao,
W. L. Pressure Induced Metallization with Absence of Structural
Transition in Layered Molybdenum Diselenide. *Nat. Commun.* 2015,
6, 7312.

(20) Liu, B.; Han, Y.; Gao, C.; Ma, Y.; Peng, G.; Wu, B.; Liu, C.;
Wang, Y.; Hu, T.; Cui, X.; Ren, W.; Li, Y.; Su, N.; Liu, H.; Zou, G.
Pressure Induced Semiconductor-Semimetal Transition in WSe2. *J. Phys. Chem. C* 2010, *114*, 14251–14254.

590 (21) Wang, X.; Chen, X.; Zhou, Y.; Park, C.; An, C.; Zhou, Y.; 591 Zhang, R.; Gu, C.; Yang, W.; Yang, Z. Pressure-Induced Iso-Structural 592 Phase Transition and Metallization in WSe2. *Sci. Rep.* **2017**, *7*, 46694.

593 (22) Qi, Y.; Naumov, P. G.; Ali, M. N.; Rajamathi, C. R.; Schnelle,

594 W.; Barkalov, O.; Hanfland, M.; Wu, S.-C.; Shekhar, C.; Sun, Y.; Süß,

595 V.; Schmidt, M.; Schwarz, U.; Pippel, E.; Werner, P.; Hillebrand, R.; 596 Förster, T.; Kampert, E.; Parkin, S.; Cava, R. J.; Felser, C.; Yan, B.; 597 Medvedev, S. A. Superconductivity in Weyl Semimetal Candidate

598 MoTe2. Nat. Commun. 2016, 7, 11038.

599 (23) Zhao, X.-M.; Liu, H.-Y.; Goncharov, A. F.; Zhao, Z.-W.; 600 Struzhkin, V. V.; Mao, H.-K.; Gavriliuk, A. G.; Chen, X.-J. Pressure 601 Effect on the Electronic, Structural, and Vibrational Properties of 602 Layered 2H-MoTe2. *Phys. Rev. B* **2019**, *99*, No. 024111.

603 (24) Akinwande, D.; Petrone, N.; Hone, J. Two-Dimensional 604 Flexible Nanoelectronics. *Nat. Commun.* **2014**, *5*, 5678.

605 (25) Peña-Álvarez, M.; del Corro, E.; Morales-García, A.; Kavan, L.;
606 Kalbac, M.; Frank, O. Single Layer Molybdenum Disulfide Under
607 Direct Out-of-Plane Compression: Low-Stress Band-Gap Engineer608 ing. Nano Lett. 2015, 15, 3139–3146.

609 (26) García, Á. M.; del Corro, E.; Kalbac, M.; Frank, O. Tuning the
610 Electronic Properties of Monolayer and Bilayer Transition Metal
611 Dichalcogenide Compounds under Direct Out-of-Plane Compression.

612 Phys. Chem. Chem. Phys. 2017, 19, 13333-13340.

613 (27) Ritschel, T.; Trinckauf, J.; Garbarino, G.; Hanfland, M.; 614 Zimmermann, M. V.; Berger, H.; Büchner, B.; Geck, J. Pressure 615 Dependence of the Charge Density Wave in 1T-TaS2 and its Relation 616 to Superconductivity. *Phys. Rev. B* **2013**, *87*, 125135.

617 (28) Dutta, U.; Malavi, P. S.; Sahoo, S.; Joseph, B.; Karmakar, S. 618 Pressure-Induced Superconductivity in Semimetallic1T-TiTe2and Its

619 Persistence Upon Decompression. Phys. Rev. B 2018, 97, 060503.

(29) Sahoo, S.; Dutta, U.; Harnagea, L.; Sood, A. K.; Karmakar, S. 620 Pressure-Induced Suppression of Charge Density Wave and 621 Emergence of Superconductivity in1T-VSe2. *Phys. Rev. B* 2020, 622 *101*, No. 014514. 623

(30) Kang, M.; Rathi, S.; Lee, I.; Lim, D.; Wang, J.; Kahn, M. A.; 624 Kim, G.-H. Electrical Characterization of Multilayer HfSe2 Field- 625 Effect Transistors on SiO2 Substrates. *Appl. Phys. Lett.* **2015**, 97, 626 143108. 627

(31) Yin, L.; Xu, K.; Wen, Y.; Wang, Z.; Huang, Y.; Wang, F.; Shifa, 628 T. A.; Cheng, R.; Ma, H.; He, J. Ultrafast and Ultrasensitive 629 Phototransistors Based on Few-Layered HfSe2. *Appl. Phys. Lett.* **2016**, 630 *109*, 213105. 631

(32) Yue, R.; Barton, A. T.; Zhu, H.; Azcatl, A.; Pena, L. F.; Wang, J.; 632 Peng, X.; Lu, N.; Cheng, X.; Addou, R.; McDonnell, S.; Colombo, L.; 633 Hsu, J. W. P.; Kim, J.; Kim, M. J.; Wallace, R. M.; Hinkle, C. L. HfSe2 634 Thin Films: 2D Transition Metal Dichalcogenides Grown by 635 Molecular Beam Epitaxy. ACS Nano **2015**, *9*, 474–480. 636

(33) Mleczko, M. J.; Zhang, C.; Lee, H. R.; Kuo, H.-H.; Magyari- 637 Köpe, B.; Moore, R. G.; Shen, Z.-X.; Fisher, I. R.; Nishi, Y.; Pop, E. 638 HfSe2 and ZrSe2: Two-Dimensional Semiconductors with Native 639 High- κ Oxides. *Sci. Adv.* **201**7, *3*, No. e1700481. 640

(34) Prescher, C.; Prakapenka, V. B. DIOPTAS: a Program for 641 Reduction of Two-dimensional X-ray Diffraction Data and Data 642 Exploration. *High Press. Res.* **2015**, 35, 223–230. 643

(35) Rodríguez-Carvajal, J. Recent Advances in Magnetic Structure 644 Determination by Neutron Powder Diffraction. *Physica B: Condens.* 645 *Matter* 1993, 192, 55–69. 646

(36) Kresse, G.; Furthmüller, J. Efficient Iterative Schemes for ab 647 initio Total-Energy Calculations Using a Plane-Wave Basis Set. *Phys.* 648 *Rev. B* 1996, 54, 11169–11186. 649

(37) Perdew, J. P.; Burke, K.; Ernzerhof, M. Generalized Gradient 650 Approximation Made Simple. *Phys. Rev. Lett.* **1996**, 77, 3865–3868. 651

(38) Grimme, S.; Antony, J.; Ehrlich, S.; Krieg, H. A Consistent and 652 Accurate ab initio Parametrization of Density Functional Dispersion 653 Correction (DFT-D) for the 94 Elements H-Pu. *J. Chem. Phys.* **2010**, 654 *132*, 154104. 655

(39) Grimme, S.; Ehrlich, S.; Goerigk, L. Effect of the Damping 656 Function in Dispersion Corrected Density Functional Theory. J. 657 Comput. Chem. 2011, 32, 1456–1465. 658

(40) Morales-García, Á.; Illas, F. Comprehensive Analysis of the 659 Influence of Dispersion on Group-14 Rutile-type Solids. *Phys. Rev.* 660 *Mater.* **2020**, *4*, No. 073601. 661

(41) Blöchl, P. E. Projector Augmented-Wave Method. *Phys. Rev. B.* 662 **1994**, *50*, 17953–17979. 663

(42) Kresse, G.; Joubert, D. From Ultrasoft Pseudopotentials to the 664 Projector Augmented-Wave Method. *Phys. Rev. B* **1999**, *59*, 1758–665 1775. 666

(43) Monkhorst, H. J.; Pack, J. D. Special Points for Brillouin-Zone 667 Integrations. *Phys. Rev. B* **1976**, *13*, 5188–5192. 668

(44) Morales-García, Á.; Valero, R.; Illas, F. An Empirical, yet 669 Practical Way To Predict the Band Gap in Solids By Using Density 670 Functional Band Structure Calculations. J. Phys. Chem. C 2017, 121, 671 18862–18866. 672

(45) Krukau, A. V.; Vydrov, O. A.; Izmaylov, A. F.; Scuseria, G. E. 673 Influence of the Exchange Screening Parameter on the Performance of 674 Screened Hybrid Functionals. *J. Chem. Phys.* **2006**, *125*, 224106. 675

(46) Rajaji, V.; Dutta, U.; Sreeparvathy, P. C.; Sarma, S. C.; Sorb, Y. 676 A.; Joseph, B.; Sahoo, S.; Peter, S. C.; Kanchana, V.; Narayana, C. 677 Structural, Vibrational, and Electrical Properties of 1T-TiTe2 Under 678 Hydrostatic Pressure: Experiments and Theory. *Phys. Rev. B* 2018, 97, 679 No. 085107. 680

(47) Zhou, Y.; Chen, C.; Zhou, Y.; Chen, X.; Gu, C.; An, C.; Zhang, 681 B.; Yuan, Y.; Wu, H.; Zhang, R.; Zhang, L.; Zhu, X.; Yang, X.; Yang, Z. 682 Pressure-induced Evolution of Structural and Electronic Properties in 683 TiTe2. *Phys. Rev. B* **2019**, *99*, 125104. 684

(48) Hu, K.; Lian, J.; Zhu, L.; Chen, Q.; Xie, S.-Y. Prediction of 685
Fe2P-type TiTe2 Under Pressure. *Phys. Rev. B* 2020, *101*, 134109. 686
(49) Mora-Fonz, D.; Schön, J. C.; Prehl, J.; Woodley, S. M.; Catlow, 687
C. R. A.; Shluger, A. L.; Sokol, A. A. Real and Virtual Polymorphism 688

689 of Titanium Selenide with Robust Interatomic Potentials. J. Mater. 690 Chem. A **2020**, *8*, 14054.

- 691 (50) Yu, F.; Sun, J.-X.; Zhou, Y.-H. The High-pressure Phase 692 Transition of TiS2 from First-principles Calculations. *Solid State Sci.* 693 **2010**, *12*, 1786.
- 694 (51) Léger, J. M.; Pereira, A. S.; Haines, J.; Jobic, S.; Brec, R. Phase

695 Transformations of Polymeric CdI2-type IrTe2 under high pressure. J. 696 Phys. Chem. Solids **2000**, 61, 27.

697 (52) Birch, F. Finite Elastic Strain of Cubic Crystals. *Phys. Rev.* **1947**, 698 71, 809–824.

699 (53) Murnaghan, F. D. The Compressibility of Media under 700 Extreme Pressures. *Proc. Nat. Acad. Sci.* **1944**, *30*, 244–247.

701 (54) Zhang, X.; Tan, Q. H.; Wu, J.-B.; Shi, W.; Tan, P. H. Review on 702 the Raman Spectroscopy of Different Types of Layered Materials. 703 *Nanoscale* **2016**, *8*, 6435–6450.

704 (55) Zhang, X.; Qiao, X.-F.; Shi, W.; Wu, J. B.; Jiang, D. S.; Tan, P.

705 H. Phonon and Raman Scattering of Two-Dimensional Transition 706 Metal Dichalcogenides from Monolayer, Multilayer to Bulk Material. 707 *Chem. Soc. Rev.* **2015**, *44*, 2757–2785.

708 (56) Cingolani, A.; Lugarà, M.; Lévy, F. Resonance Raman 709 Scattering in HfSe2 and HfS2. *Phys. Scr.* **1988**, *37*, 389–391.

710 (57) Katkanant, V.; Kirby, R. D. Mixed-Crystal Lattice Dynamics of 711 HfxTi1-xSe2. *Phys. Rev. B* **1989**, *40*, 1152–1158.

712 (58) Greenaway, D. L.; Nitsche, R. Preparation and Optical 713 Properties of Group IV-VI2 Chalcogenides Having the CdI2 714 Structure. J. Phys. Chem. **1965**, 26, 1445–1458.

715 (59) Gaiser, C.; Zandt, T.; Krapf, A.; Serverin, R.; Janowitz, C.; 716 Manzke, R. Band-Gap Engineering with HfSxSe2-x. *Phys. Rev. B* **2004**, 717 69, No. 075205.

718 (60) Guzman, D. M.; Strachan, A. Role of Strain on Electronic and 719 Mechanical Response of Semiconducting Transition-Metal Dichalco-720 genide Monolayers: An ab-initio Study. *J. Appl. Phys.* **2014**, *115*, 721 243701.