



RESEARCH PAPER

# Ordovician volcanic rocks record rifting, Variscan metamorphism and gold mineralization processes (Truchas Syncline, NW Iberia, Spain)

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## Abstract

Ordovician volcanic rocks outcrop in several locations of the NW Iberian Variscan belt. Their composition is mainly basaltic (with less acid types) and occur as volcanic-volcanoclastic layers within a shale-slate succession. This work focuses on volcanic and related rocks within a prominent Variscan structure, the Truchas Syncline. We studied field relations, petrography, mineralogy, geochemistry and conducted thermodynamic modelling to review the petrogenesis and establish the evolution of these volcanic rocks classified as within-plate alkaline basalts (high Ti/Y, Nb/Y and Nb/Yb). Crustal contamination is absent given the elevated Nb/La ratio (1–1.5). These features indicate low melting degrees of the upper mantle and a continental rifting environment. The finding of Ordovician orthid brachiopods in some of the volcanoclastic rocks suggests a shallow marine environment for the volcanic deposition. Variscan metamorphism occurred at lower greenschist conditions with chlorite-temperatures of  $374 \pm 6$  °C. Quartz + carbonate veins indicate that H<sub>2</sub>O–CO<sub>2</sub> metamorphic fluids traversed some volcanic rocks, reacting with Ca–Fe–Mg phases to produce carbonates (Mg–calcite–Fe–dolomite). For this event, T-XCO<sub>2</sub> modelling indicates temperatures below 350–360 °C and fluid XCO<sub>2</sub> between 0.10 and 0.45. Such fluids can be important carriers of Au and might explain gold deposits in adjacent quartzites. Metasomatic shales (Fe-chlorite + quartz) outcrop nearby and were derived from a mixed protholith of shales and minor volcanic components. Its geochemistry shows Fe enrichment and high peraluminosity. Variscan deformation further modified its geochemistry causing Si-depletions and relative increases of other elements (K, Na, Ti, Al, Rb, Sr, Ba and LREE) in shear zones domains.

**Keywords** Volcanic rocks · Metamorphism · Ordovician · Variscan · Truchas Syncline · Luarca formation

## Resumen

Rocas volcánicas Ordovícicas afloran en el Macizo Varisco del noroeste de la Península Ibérica. Su composición es basáltica (con escasos términos ácidos) y forman niveles volcánicos y volcanoclásticos dentro de una sucesión de pizarras Ordovícicas. En este trabajo se estudian estas rocas en el Sinclinal de Truchas, una importante estructura Varisca. Se investigan las relaciones de campo, petrografía, mineralogía, geoquímica y se realiza un modelo termodinámico para revisar la petrogénesis de estas rocas volcánicas clasificadas como basaltos alcalinos intra-placa con elevadas relaciones de Ti/Y, Nb/Y y Nb/Yb. No se observan indicios de contaminación cortical, dadas las elevadas relaciones de Nb/La observadas (1–1.5). Esta geoquímica indica bajos grados de fusión del manto superior y un ambiente tectónico de rifting continental. La presencia de restos de braquiópodos órtidos del Ordovícico en algunas rocas volcanoclásticas sugiere un ambiente marino poco profundo para su depósito. El metamorfismo varisco se produjo en condiciones de esquistos verdes con temperaturas de clorita de  $374 \pm 6$  °C.

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35 Diversas venas de cuarzo y carbonatos observadas representan fluidos metamórficos de H<sub>2</sub>O–CO<sub>2</sub> que atravesaron algunas  
 36 rocas volcánicas, reaccionando sus minerales de Ca–Fe–Mg para producir carbonatos (Mg–calcita–Fe–dolomita). Para este  
 37 evento, el modelo de T–XCO<sub>2</sub> realizado indica temperaturas por debajo de 350–360 °C y XCO<sub>2</sub> del fluido entre 0.10 y 0.45.  
 38 Estos fluidos pueden ser portadores de oro en solución y podrían explicar mineralizaciones descritas en las cuarcitas adya-  
 39 centes. Varios niveles de pizarras metasomáticas (formadas por clorita rica en hierro y cuarzo) afloran en zonas próximas.  
 40 Derivan de un protolito mixto de pizarras y componentes volcánicos menores. Su geoquímica muestra enriquecimiento en  
 41 hierro y elevada peraluminosidad. La deformación varisca modificó adicionalmente su geoquímica causando empobreci-  
 42 mientos de sílice e incrementos relativos de otros elementos en zonas de cizalla (K, Na, Ti, Al, Rb, Sr, Ba y LREE).

43 **Palabras Clave** Rocas volcánicas · Metamorfismo · Ordovícico · Varisco · Sinclinal de Truchas · Formación Luarca

## 44 1 Introduction

45 **Abstract** Ancient volcanic and volcanoclastic rocks are important  
 46 in earth science research because they can contain a rich  
 47 record of past geological conditions: the nature of their  
 48 mantle/crustal sources and the composition of the crust tra-  
 49 versed if contamination processes happened. Even crustal  
 50 and lithospheric thicknesses estimations are possible from  
 51 the study of volcanic rocks (Liu et al. 2016 and references  
 52 therein). Furthermore, they record the type of emplacement  
 53 conditions, possible contact metamorphism effects, surficial  
 54 environmental conditions and type of alteration processes.  
 55 Volcanic rocks of basic to intermediate composition are  
 56 particularly susceptible to metamorphic and hydrothermal  
 57 processes affecting their geological surroundings (Bucher  
 58 and Grapes 2011). Their ages can also be calculated by dif-  
 59 ferent radiometric methods (K–Ar, Ar–Ar, Sm–Nd) provided  
 60 alteration has not been very intense.

61 Lower Paleozoic volcanic (±volcanoclastic) rocks in the  
 62 NW Iberian Variscan belt have been studied to infer the  
 63 nature of the mantle source and emplacement conditions  
 64 prior the Variscan orogenic event (Valverde-Vaquero 1992;  
 65 Gallastegui et al. 1992; Suárez et al. 1993; Villa et al. 2004;  
 66 Brendan Murphy et al. 2008 among others). These volcanic  
 67 rocks occur in different geological domains of NW Spain  
 68 and usually record their emplacement conditions and sub-  
 69 sequent evolution through alteration, metasomatism, Vari-  
 70 scan metamorphism-deformation and gold mineralization in  
 71 nearby host rocks (Villa et al. 2004).

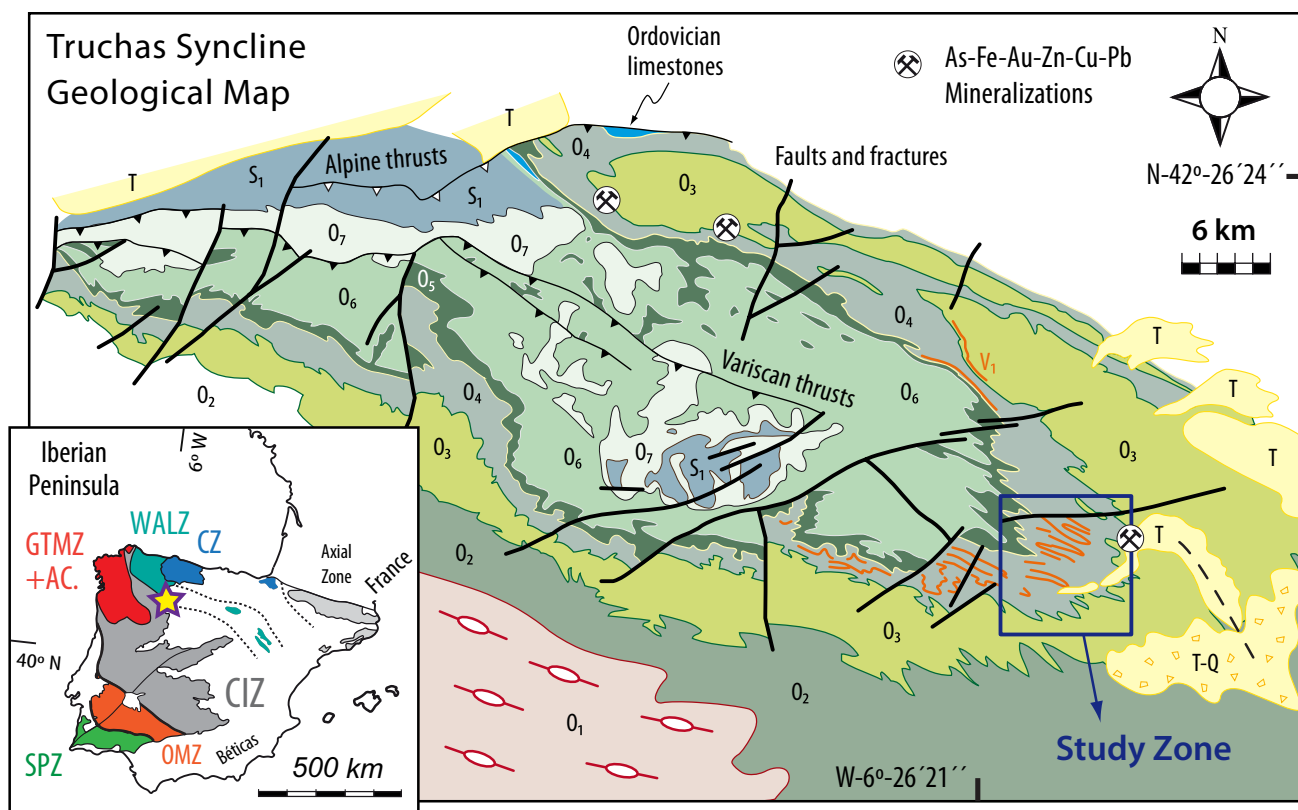
72 The Variscan Truchas syncline (Fig. 1; Suárez et al. 1994;  
 73 Fernández-Lozano, 2012; Rodríguez Fernández et al. 2015)  
 74 is a prominent Carboniferous structure located in the north-  
 75 ern part of the Central Iberian Zone (CIZ). It is composed  
 76 of lower Paleozoic shale-sandstone-quartzite successions  
 77 affected by folds, foliations and a low metamorphic grade  
 78 (greenschists) attained during the Variscan orogenic event  
 79 (≈ 320–350 Ma). Mafic ± felsic volcanic and volcanoclastic  
 80 rocks of Ordovician age occur as sills and layers within the  
 81 shale formations. Furthermore, late Variscan gold miner-  
 82 alization appear in quartz veins near the contact between

Ordovician volcanic rocks, shales and quartzites (Gómez-  
 Fernández et al. 2012).

The present work focuses on the field relations, fossil  
 occurrences, petrography, mineralogy and geochemistry  
 of the SE Truchas syncline volcanic and related rocks. Our  
 aim is to understand its origin, emplacement environment,  
 hydrothermal alteration, metamorphism, metasomatism,  
 deformation and possible relations with nearby gold min-  
 eralization. All these processes have played a role in con-  
 figuring the current features in these rocks, causing in some  
 cases, deviations from the expected whole rock composi-  
 tion (González-Menéndez et al. 2019). This study reinforces  
 the rifting hypothesis for the origin of the studied basaltic  
 volcanic rocks. Some of them are interpreted as metasoma-  
 tic, derived from shales ± sandstones, mixed with a minor  
 volcanic component and affected by Fe-metasomatism and  
 SiO<sub>2</sub>-depletion processes. H<sub>2</sub>O–CO<sub>2</sub> late-stage fluids (post-  
 peak metamorphic), known to be candidates for gold trans-  
 port in orogenic settings (Kesler 2005; Yardley and Bodnar  
 2014), left a key record in the form of carbonation of some  
 of the mafic volcanic rocks studied.

## 2 Geological background and sampling site description

The Truchas syncline region has a very complete and rep-  
 resentative record of Ordovician sedimentary rocks (shales,  
 sandstones, quartzites, ± limestones) including abundant  
 volcanic and volcanoclastic rocks. The basal part of the Tru-  
 chas syncline is formed by the Ollo de Sapo metavolcanic  
 orthogneiss (Fig. 2) (Díez Montes 2007), cropping out to  
 the south and to the east of the studied zone (Fig. 1). The  
 overlying sedimentary formations are: Capas de los Montes  
 (Lower Ordovician quartzites, microconglomerates, black  
 shales and sandstones), Cuarcita Armoricana (quartzites,  
 Lower to Middle Ordovician), Capas de Transición (shales  
 and sandy shales), Luarca Formation (Middle Ordovician  
 black shales-slates), interlayered mafic/felsic metavolcanic  
 rocks, the Casaio Formation (Upper Ordovician quartzites,



**Fig. 1** Geological map of the Truchas Syncline region, NW Spain (modified after Suárez et al. 1994). *O1* Ollo de Sapo orthogneisses, *O2* Capas de los Montes Formation: schists, quartzites and sandstones, *O3* Cuaucita Armoricana and Capas de Transición formations, *O4* Luarca Formation, shales-slates, *V1* volcanic and volcanoclastics layers and sills, *O5* Casaio Formation, shales ± quartzites, *O6* Rozadais Formation, shales, *O7* Losadilla Formation, shales ± sandstones. *S1* Llagarinos Formation, ampelitic shales, *T* Tertiary, conglomerates and sands, *CZ* Cantabrian Zone, *WALZ* West Asturian-Leonese Zone, *GTMZ* Galicia-Trás-os-Montes Zone, *CIZ* Central Iberian Zone, *OMZ* Ossa Morena Zone, *SPZ* South Portuguese Zone

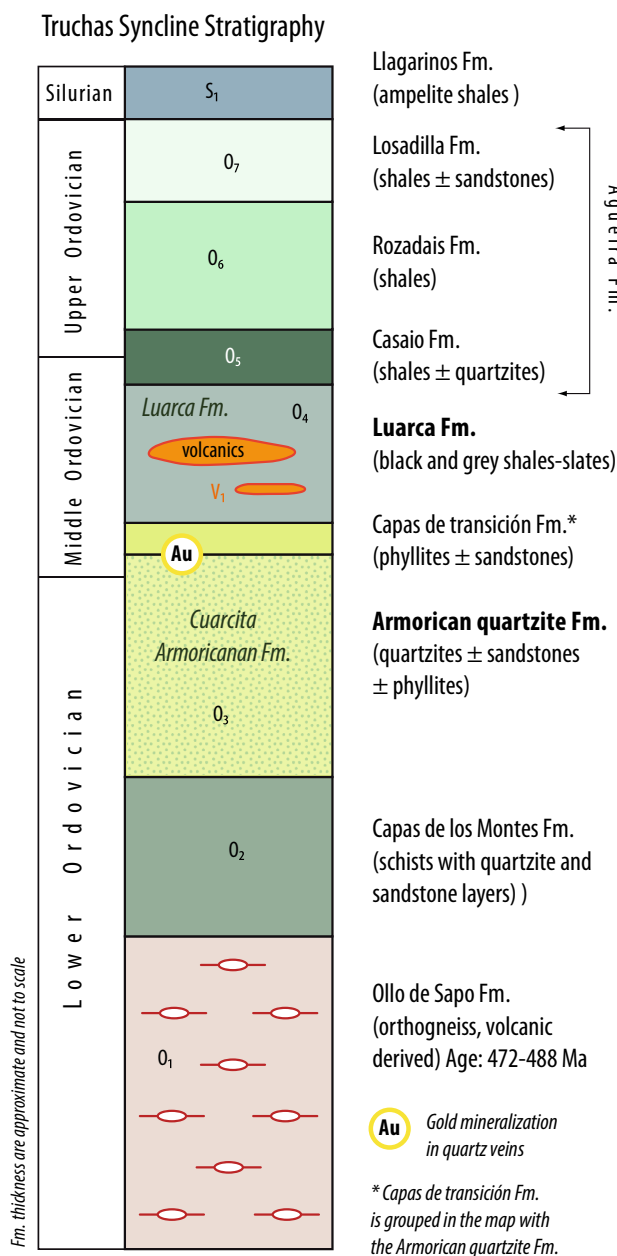
120 shales and rare carbonate beds), the Rozadais Formation  
 121 (Upper Ordovician shales), the Losadilla Formation (Upper  
 122 Ordovician shales and sandstones) and the Llagarinos For-  
 123 mation (Silurian ampelitic black shales ± sandstones).

124 The volcanic rocks are interlayered with shales (Luarca  
 125 Fm) and were described in the geological maps of this area  
 126 (Suárez et al. 1994; Villar Alonso et al. 2019 and references  
 127 therein). The shales have been attributed to a Middle Ordovi-  
 128 cian age by paleontological studies by Gutiérrez et al. (1999;  
 129 2002). Regarding the volcanic rocks, petrological-geochemi-  
 130 cal studies of these and other outcrops from the Central-Iber-  
 131 ian Zone (CIZ), the West-Asturian Leonese Zone (WALZ)  
 132 and the Cantabrian Zone (CZ) have been done by different  
 133 authors (Valverde-Vaquero 1992; Gallastegui et al. 1992;  
 134 Suárez et al. 1993; Villa et al. 2004; Brendan Murphy et al.  
 135 2008). These studies suggest a crustal origin for the felsic  
 136 types, an enriched mantle source for the mafic ones and an  
 137 intraplate rifting environment for the intrusion and emplace-  
 138 ment of all these volcanic rocks.

139 The geodynamic setting for the Ordovician sedimenta-  
 140 tion and the felsic magmatism in central and NW Iberia has

141 been discussed by Díez Montes 2007 and references therein. 141  
 142 Valverde-Vaquero and Dunning (2000) proposed a back-arc 142  
 143 setting to explain the felsic magmatism and volcanism repre- 143  
 144 sented by the Ollo de Sapo formation. Recent interpretations 144  
 145 propose different hypotheses: an environment evolving from 145  
 146 a compressive geodynamic setting to a passive one (Vil- 146  
 147 laseca et al. 2016) vs. a purely extensional regime (Montero 147  
 148 et al. 2017). 148

149 In the present work, we study and review the previous 149  
 150 mapping of volcanic rocks at the southeast domain of the 150  
 151 Truchas syncline (Fig. 3). The volcanic layers were affected 151  
 152 by recumbent/inclined, meter-scale folds, apparently south 152  
 153 vergent but coherent with the main fold geometry of the Tru- 153  
 154 chas syncline (Suárez et al. 1994; Rodríguez Fernández et al. 154  
 155 2015). Some of the volcanoclastic rocks preserve a stratifica- 155  
 156 tion (Fig. 4a) that records volcanoclastic deposition. In some 156  
 157 outcrops, scoriaceous layers bordering the main volcanic 157  
 158 masses appears (Fig. 4b, c) attesting to the presence of lava 158  
 159 bodies, deposited above the shales, and cooling rapidly in 159  
 160 their upper borders (samples C-1B, CF-2, and M3 in Fig. 3). 160  
 161 Samples IGS-39 and IGS-40 are volcanic rocks located 2 km 161



**Fig. 2** Simplified stratigraphy of the Truchas Syncline region, NW Spain (modified after Suárez et al. 1994). The Upper Ordovician formations, Casaio, Rozadais and Losadilla, formed by shales ± sandstones, are grouped in other northern geological settings, into the Agüeira Formation

to the south of the study zone. Other sets of rocks have a more difficult assignment having no clear volcanic features, lacking top scoriaceous layers and showing a high foliation density in their contacts with the shales (samples CF-3, C-5, C-5R, C-6, C-6RA, C-6RB and C-7 in Fig. 3). We named these rocks metasomatic shales due to significant geochemical differences when compared to volcanic or volcanoclastic rocks and having more affinity with shale/slate compositions

(Ward and Gómez-Fernández 2003; Gómez-Fernández et al. 2009).

### 3 Fossil occurrences in the volcanoclastic rocks

The existence of a diverse fossil record in the Luarca Formation (equinoderms, bryozoans, bivalves, rostroconchs, cephalopods, gastropods, brachiopods, ostracods, hyolithids, graptolites and trilobites) has been observed in northwestern Spain (CIZ, WALZ and CZ) where these materials extensively outcrop (Gutiérrez-Marco et al. 1999 and references therein). Nevertheless, the presence of fossil fauna within the volcanoclastic levels interlayered in the Luarca Formation is scarce. Pérez-Estaún (1974; 1978) was the first who reported the occurrence of *Redonia* sp. (shells of bivalves determined by prof. C. Babin) in volcanoclastic layers close to the base of the Luarca Formation (Valdavidó, southern Truchas Syncline). Matas and Velando (1982) also reported the presence of *Redonia* in the volcanoclastic levels of the studied area at northern Cunas. *Redonia* findings allowed assigning an Ordovician age to these levels, with no further precisions. Subsequently, Babin and Gutiérrez-Marco (1991) reassigned as *Redoniades hayesi* the previous *Redonia* sp. bivalves. New fossil material, *Tolmachovia* n. sp. and *Porambonites?* sp., were described from the first volcanoclastic levels referenced (Gutiérrez-Marco et al. 1999) and an accurately age was established: Oretanian.

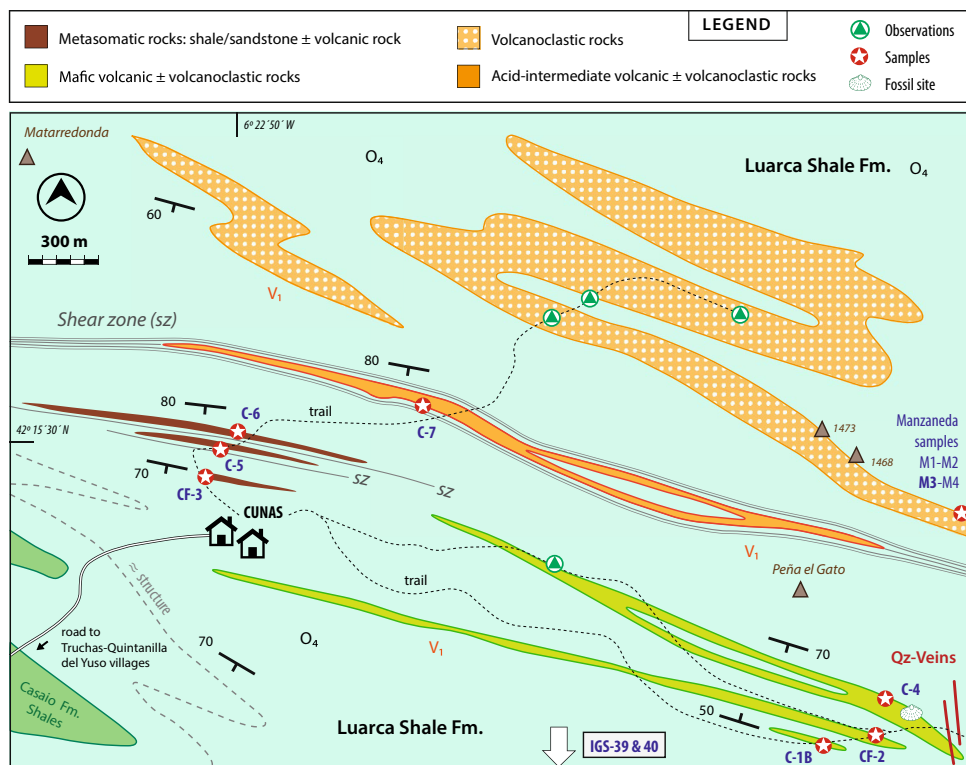
In the Real syncline (Mondoñedo Nappe, WALZ) there are also thin volcanic intercalations at the base of Luarca Shales Formation where Marcos et al. (1980) cited fragments of indeterminate shells. Emig and Gutiérrez-Marco (1997) assigned these fossils to lingulid bivalves concentrated in a single horizon.

At the Cabo Peñas and Cabo Vidrias (CZ) volcanoclastic rocks interlayered with sandstones, shales and slates, form the lower Sect. (400 m thick) of the of the Castro Formation that conformably overlies the Luarca Formation. The fossil record here is composed of *Mecwanella vulcanica*, *Hesperinia asturica*, *Ectillaenus giganteus*, *Pinaceocladichnus bulbosus*, *Ogmoopsis?* sp., *Asaphina* gen. indet., *Bryozoa* gen. indet. and *Pelmatozoa* gen. indet. These levels have been reassigned as late Dobrotivian age (Villas et al. 1989; Gutiérrez-Marco et al. 1999). Paleoenvironmental conditions related to these volcanoclastic horizons indicate calm and locally warm water around the volcanic emission sites, episodically colonized by opportunistic communities of bryozoans, echinoderms and brachiopods (Gutiérrez-Marco et al. 1999).

A new fossil record is reported here from the studied area. Volcanoclastic rocks of Cunas C-4 site (Fig. 3) preserve outer molds of orthid brachiopod valves (Fig. 4d).



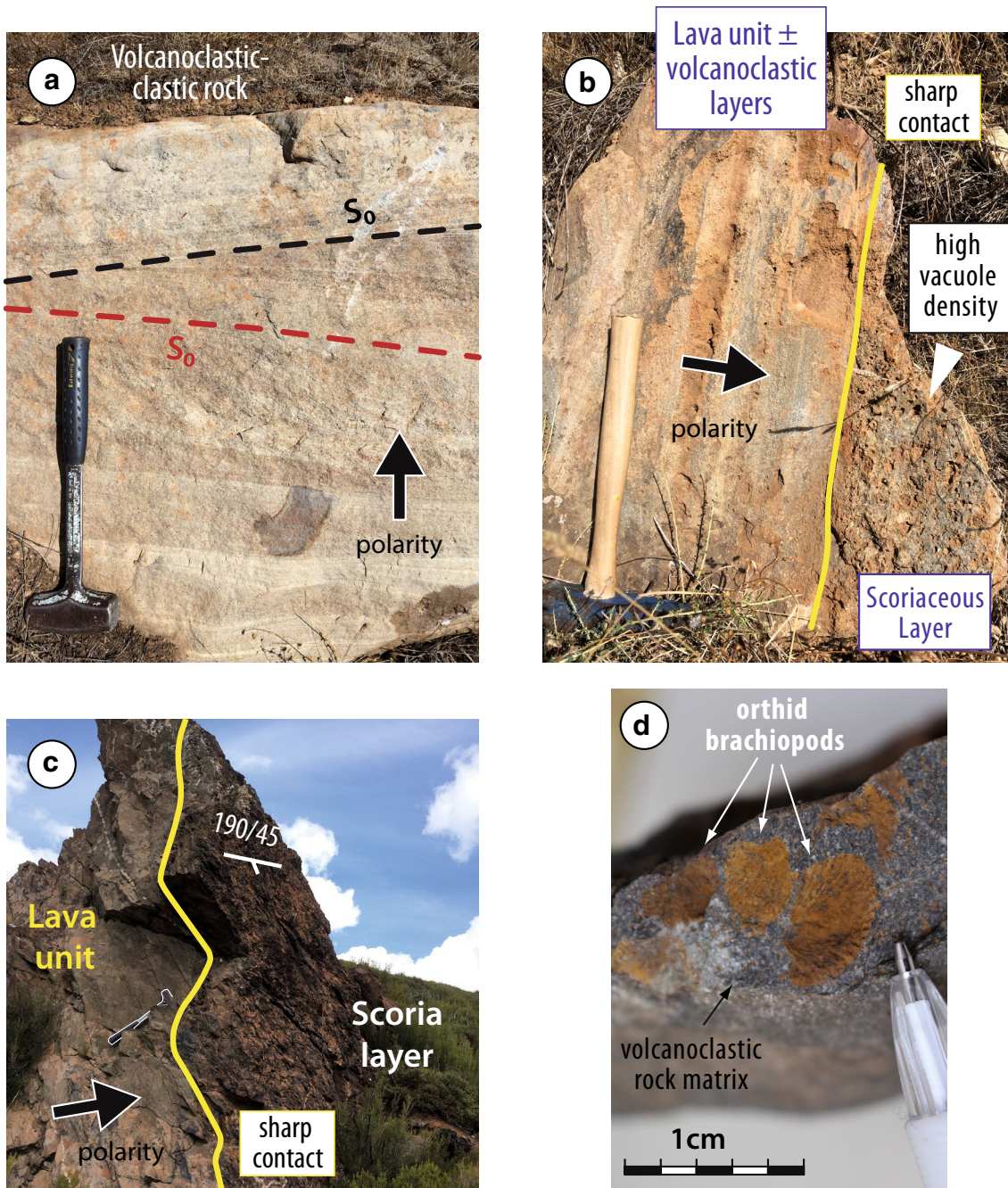
**Fig. 3** Geological map of the studied zone (modified after Suárez et al. 1994; Villar Alonso et al. 2019). Sampling is shown with the symbol stars. Locations C-5 and C-6 include sets of samples in order to record the variation in foliation intensity (C-5: C-5, C-5R; C-6: C-6, C-6RA, C-6RB). Bold numbers correspond to samples where EMPA and/or bulk rock analyses was performed



220 Unfortunately, the preservation of samples is very poor for  
 221 an accurate taxonomical determination. They have been  
 222 collected to be housed in Geominero Museum under the  
 223 numbers from MGM-81520 to MGM-81600. Valves, both  
 224 dorsal and ventral, are disarticulated; most of them show  
 225 no evidence of sorting or preferred orientation. They have  
 226 not suffered significant breakage and therefore did not suffer  
 227 considerable transportation. Conversely, the emission of  
 228 volcanic materials implies a very high ambient temperature  
 229 and survival in the site emission probably was not possible.  
 230 Thus, the depositional ambient could have been a low-energy  
 231 media with fine-medium sand and small size basalt debris  
 232 supplied by nearby volcanic eruptions. Villas et al. (1989)  
 233 suggest that a low rate of sedimentation and long periods  
 234 before burial, subsequent to death, enhanced the disarticulation  
 235 of shells and infestation by borers. These conditions  
 236 could have been similar in the studied area. This is supported  
 237 by the occurrence of *Redonia deshayesi* and *Tolmachovia* sp.  
 238 in these volcanoclastic levels. They are usually infaunal filter  
 239 feeders but some specimens of spanish *Redonia deshayesi*  
 240 show a bryozoan incrustation, suggesting that a part of the  
 241 shell projected above the seabed (Polechová 2016) which  
 242 normally requires a low rate of sedimentation. *Tolmachovia*  
 243 environment conditions are interpreted to be quiet and sub-  
 244 littoral marine platform (Gutiérrez-Marco 1997).

## 4 Methods

245 Field geology and sampling was conducted in the south-  
 246 eastern part of the syncline where an important outcrop  
 247 area of volcanic rocks is mapped in the region. Samples  
 248 were taken from different volcanic/volcanoclastic units in  
 249 large enough amounts (5–10 kg) to ensure chemical and  
 250 mineralogical homogeneity of the rock samples. Thin sections  
 251 were made for each sample for petrography studies  
 252 under the optical microscope, SEM (scanning electronic  
 253 microscope) and EMPA (electron micro-probe analysis)  
 254 investigations. SEM was done at Leon University (Spain)  
 255 using a JEOL JSM-6480 scanning electron microscope  
 256 equipped with an Oxford D6679 EDS detector. EMPA was  
 257 done at Oviedo University (Spain) using a Cameca SX-100  
 258 electron microprobe operated at 15 keV accelerating volt-  
 259 age, 15 nA beam current and and 2  $\mu$ m beam size. Ortho-  
 260 clase (Si), wollastonite (Ca), MnTi (Ti, Mn), magnetite  
 261 (Fe), albite (Na), Al<sub>2</sub>O<sub>3</sub> modified (Al), cromite (Cr), NiO  
 262 (Ni), MgO (Mg) and apatite (P) were used as standards  
 263 for determination of the respective elements in brackets.  
 264 Most of the bulk rock analysis was performed by ICP-AES  
 265 (Inductively Coupled Atomic Emission Spectroscopy)  
 266 for major elements and ICP-MS (Inductively Coupled  
 267



**Fig. 4** Field relations of the studied volcanic/volcanoclastic rocks. **a** Cross bedding in a mixed volcanic-sandstone rock. **b** Scoriaceous layer developed in the border of a volcanic rock. **c** Volcanic rock

interpreted as a lava flow shielded in its upper border by a scoria layer. **d** Shell molds of Ordovician orthid brachiopods within a volcanoclastic layer (C-4 in Fig. 3.)

268 Mass Spectrometry) for trace and rare earth elements.  
 269 The analyses were carried out at the ALS Global laboratory  
 270 in Ireland (ME-ICP06 and ME-MS81 procedures).  
 271 A simplified summary of these methods is given below.  
 272 For the major oxides, a milled sample (0.1 g; <0.05 mm  
 273 grain size) is added to lithium metaborate (35.3%)/lithium  
 274 tetraborate (64.7%) flux, mixed well and fused in a furnace  
 275 at 1000 °C. The resulting melt is then cooled and dissolved

276 in 100 ml of 4% nitric acid + 2% hydrochloric acid. This  
 277 solution is then analyzed by ICP-AES and the results are  
 278 corrected for spectral inter-element interferences. For  
 279 the trace and rare earth elements, a milled sample (0.1 g;  
 280 <0.05 mm grain size) is added to lithium metaborate  
 281 (35.3%) / lithium tetraborate (64.7%) flux, mixed well and  
 282 fused in a furnace at 1025 °C. The resulting melt is then  
 283 cooled and dissolved in an acid mixture containing nitric,



hydrochloric and hydrofluoric acids. This solution is then analyzed by inductively coupled plasma–mass spectrometry. Samples IGS-39 and IGS-40 were taken and analyzed during a previous and unpublished geological study in the region by one of the authors (G. Gallastegui). Major and some trace elements of these samples were analyzed by XRF at the Technical-Scientific Services of Oviedo University (Spain) using a WD-XRF spectrometer (model 2404; PANalytical) coupled with a Rh tube. Major element analyses were performed using glass beads of powdered rocks after fusion with lithium tetraborate. Precision of the XRF technique was better than  $\pm 1\%$  relative. Trace elements were determined on pressed pellets with Elvacite. Raw data were processed using Pro-Trace-XRF PANalytical software. Other trace elements (Cs, Ga, Hf, Ta, Li, Sc) and rare earth elements (REE) were analyzed by ICP-MS following sample decomposition with lithium metaborate at the Geochronology and Geochemistry-SGIker facility of El País Vasco University/EHU (Spain) (see García de Madinabeitia et al. 2008 for additional details). All the results are shown in Tables 1, 2 and 3.

## 5 Petrography and mineralogy

### 5.1 Volcanic and volcanoclastic rocks

The volcanic and volcanoclastic rocks are characterized by the presence of plagioclase crystals, volcanic fragments and scoriaceous margins at the top of the volcanic layers. The main petrographic features are the micro-porphyrific textures, phenocrysts of plagioclase, absence of any other primary phases and presence of secondary minerals such as K-feldspar, quartz, sericite and widespread chlorite. By these criteria, samples C-1B, CF-2, C-4, C-7, M1, M2A, M3 and M4 (Fig. 3) are classified as volcanic and volcanoclastic.

The volcanic rocks (C-1B, CF-2, C-4, M3) are composed of plagioclase phenocrysts (Fig. 5a) and some rock fragments within a matrix of plagioclase + chlorite  $\pm$  quartz  $\pm$  K-feldspar  $\pm$  sericite. Rare prehnite crystals were observed in two samples. Carbonate minerals and chlorite are abundant in some rocks and scarce in others. Carbonates replace plagioclase phenocrysts, Ca–Fe–Mg phases and rock fragments (Fig. 5b). Growth of carbonate over chlorite is observed. Quartz + carbonate and pure carbonate veins

**Table 1** Chlorite analyses by EMP. Oxide contents are in wt.%. Volc.: volcanic rock

Refs.	m3a n=1	m3b n=1	m3c n=1	m3d n=1	m3e n=1	m3f n=1	m3g n=1	m3h n=1	m3i n=1	m3j n=1	Pomb Avg n=12	Llamas Avg n=8
Rock	Volc	Volc	Volc	Volc	Volc	Volc	Volc	Volc	Volc	Volc	Slates	Slates
SiO <sub>2</sub>	25.448	25.779	26.574	25.816	25.947	25.553	25.848	25.979	25.425	26.246	23.71	23.69
TiO <sub>2</sub>	0.045	0.008	0.010	0.108	0.028	0.024	0.028	0.055	0.045	0.035	0.026	0.023
Al <sub>2</sub> O <sub>3</sub>	23.467	23.967	24.401	23.806	23.985	23.797	24.092	24.042	23.498	23.728	21.19	21.15
FeO	19.737	19.668	19.850	19.885	19.902	19.793	20.035	19.859	19.891	20.126	36.66	36.71
MnO	0.107	0.058	0.066	0.088	0.058	0.036	0.028	0.052	0.031	0.017	0.45	0.43
MgO	17.008	17.339	16.611	16.997	16.898	17.212	17.007	16.491	16.285	16.925	6.03	6.04
CaO	0.071	0.051	0.062	0.076	0.046	0.062	0.045	0.068	0.098	0.085	0.027	0.030
Na <sub>2</sub> O	0.009	0.028	0.000	0.056	0.035	0.000	0.000	0.030	0.000	0.002	0.012	0.014
Total	86.332	87.330	88.091	87.145	87.386	86.900	87.376	86.968	85.769	87.498	88.22	88.25
T °C	377.846	379.328	363.358	376.228	373.942	382.503	378.137	370.108	375.220	366.444	385.33	385.34
Refs.	C3a n=1	C3b n=1	C3c n=1	C3d n=1	C3e n=1	C3f n=1	C3g n=1	C3h n=1	C3i n=1	C3j n=1	C3k n=1	C3l n=1
Rock	Met	Met	Met	Met	Met	Met	Met	Met	Met	Met	Met	Met
SiO <sub>2</sub>	24.069	21.952	22.934	23.164	22.759	23.303	22.719	23.589	23.189	22.224	22.544	22.478
TiO <sub>2</sub>	0.039	2.598	0.033	0.068	0.004	0.074	0.030	0.017	0.035	0.056	0.000	0.133
Al <sub>2</sub> O <sub>3</sub>	24.171	23.207	23.819	22.719	23.389	23.833	23.798	23.764	23.321	23.662	23.925	23.681
FeO	34.543	35.157	35.743	36.477	35.812	36.635	36.467	34.974	36.427	36.754	36.389	35.473
MnO	0.067	0.109	0.083	0.130	0.071	0.083	0.079	0.078	0.120	0.141	0.116	0.082
CaO	0.017	0.032	0.039	0.031	0.008	0.018	0.033	0.033	0.021	0.025	0.002	0.015
Na <sub>2</sub> O	0.027	0.027	0.000	0.046	0.000	0.000	0.000	0.019	0.035	0.008	0.000	0.011
Total	87.793	88.248	87.798	88.261	87.383	89.180	88.544	88.215	88.395	88.273	88.382	87.751
T °C	386.556	454.434	415.900	407.272	416.204	414.974	428.276	402.522	411.292	440.096	431.459	430.683

Met. metasomatic shale, Pomb Pombriego, Avg Average. T °C calculated by the method of Jowett 1991

**Table 2** Carbonate analyses by EMP. Oxide contents are in wt.%. Volc.: volcanic rock

Ref	M3-1 n=1	M3-2 n=1	M3-3 n=1	M3-4 n=1	M3-5 n=1	M3-6 n=1	M3-7 n=1	M3-8 n=1	M3-9 n=1	M3-10 n=1
Rock	Volc	Volc	Volc	Volc	Volc	Volc	Volc	Volc	Volc	Volc
CaO	33.355	32.945	32.252	32.693	31.854	30.735	30.796	31.926	32.074	33.568
FeO	7.126	7.580	7.206	6.525	6.319	6.165	7.138	7.257	7.420	6.901
MgO	15.362	15.089	14.740	15.633	16.066	15.427	15.176	15.463	15.786	15.474
MnO	0.6731	0.6593	0.575	0.804	0.5129	0.6962	0.6272	0.6498	0.5252	0.8107
SiO <sub>2</sub>	0.2373	0.0692	0.0986	0.3405	0.1482	0.0847	0.078	0.0268	0.0325	0.1366
Al <sub>2</sub> O <sub>3</sub>	0.1156	0.0501	0.0166	0.2359	0.0645	0.0751	0.0848	0.0299	0.0406	0.075
P <sub>2</sub> O <sub>5</sub>	0.2164	0.2427	0.2376	0.2631	0.2077	0.169	0.2503	0.2103	0.1857	0.2411
CO <sub>2</sub>	42.531	43.087	44.756	43.297	44.599	46.422	45.692	44.075	43.833	42.582
Total	99.7034	99.962	99.978	99.971	99.8918	99.983	99.984	99.668	99.966	99.963

cut across the primary and secondary mineral assemblage in some samples (Figs. 5c & d). A SEM study of the M3 sample reveals that the carbonate phases are mainly Fe-bearing dolomite or Fe–Mg calcite, and the plagioclase phenocrysts are nearly pure albite (Na-rich). Apatite, rutile, chalcopyrite and sphalerite were identified. The textures are microphyritic with a significant difference in size between phenocrysts and matrix. Some amygdaloids occur, filled with growing crystals of quartz + feldspar, and iron oxides in the center. Slight mineral orientation/rock deformation is observed in some of the samples.

The volcanoclastic rocks (C-4, M1, M2A, M2C, M4, C-7) are composed of different volcanic fragments with similar mineralogy as the volcanic rocks: plagioclase laths in a chlorite ± quartz matrix. Accessory phases include sericite, muscovite, tourmaline, iron sulphides, and zircon. Occasionally oversized plagioclase ± quartz crystals appear. The different fragments have rounded/elliptical and lobule shapes defining a broad oriented pattern. These fragments have thin and sharp dark borders composed of iron oxides. No preferred mineral grain orientations were observed. C-7 sample is more felsic and its texture is dominated by a high foliation density.

## 5.2 Metasomatic shales

The metasomatic rocks were originally mapped as volcanic, but this affinity was dubious because they lack the field criteria previously defined. These are the samples: CF-3, C-5, C-5R, C-6, C-6RA and C-6RB (Fig. 3). Sample CF-3 has been studied with more detail (Fig. 5e,f). It is composed of quartz (≈40–45%) ± rock fragments (quartz rich, pelitic or limolitic) and a chlorite matrix (≈45–50%) that surrounds most of the other mineral phases. Minor patches of sericite also occur. Neither feldspars (plagioclase/K-feldspar) nor carbonate phases are present. The rock has a subtle subolitic orientation defined by the chlorite matrix. Quartz is not internally deformed but some crystals have their long axis

approximately parallel to the chlorite fabric. This might indicate that the deformation was mainly transferred to the more ductile chlorite matrix. A SEM study of this rock reveals that the chlorite is Fe-rich (chamosite) and the most common accessory minerals are zircon, rutile and xenotime. Samples C-5 and C-6 series are similar to CF-3, though their texture is defined by a strong foliation defined by the oriented chlorite. Foliation density changes from very high strain, at the contact with the shales, to low strain away from this contact.

## 6 Chlorite temperatures and carbonate chemistry

Chlorite is a common secondary phase appearing in all the studied rocks. In the volcanic/volcanoclastic rocks it is probably replacing Fe–Mg bearing primary minerals such as pyroxenes or amphiboles that have been completely altered. We have analyzed some of these chlorites to characterize their chemistry and to estimate their temperatures of formation. Table 1 shows the chlorite compositions (measured by EMPA) and their crystallization temperatures calculated with the method described by Jowett (1991). The correlation of between T °C and the Al<sup>4</sup> corrected term is R = 0.981. Standard deviations estimated on specific studies was between 32 and 37 °C (Jowett 1991). Chlorites from the mafic volcanic-volcanoclastic rocks are classified as Mg-chlorites. They lie in the field of clinocllore in the diagram of Zane and Weiss (1998) (Fig. 6a) and have moderate to elevated Al contents (2.85–2.90 a.f.u.). Their calculated temperatures range from 363 to 382 °C and show a mean of 374 ± 6 °C (Fig. 6b). These temperatures are very similar to those calculated for chlorites in the Ordovician host shales (Pombriego and Llamas, Fig. 6b). Chlorites occurring in the metasomatic shales and shales are classified as Fe-chlorites (Fig. 6a). Their Al content is relatively high in the metasomatic shales (3–3.14 a.f.u.) and lower in the

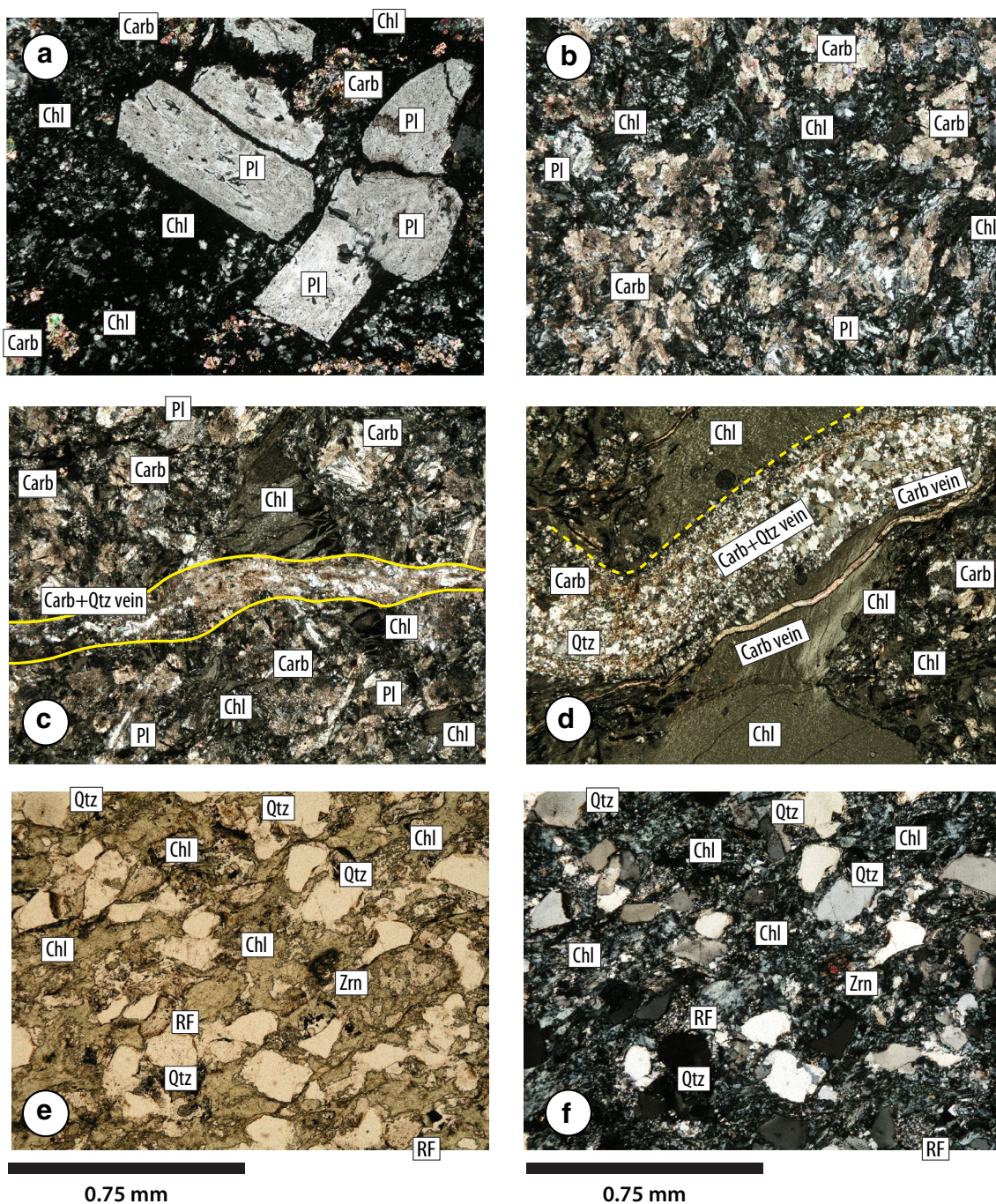


**Table 3** Bulk-rock analyses (ICP-AES, XRF, ICP-MS) of the studied samples

Ref	C-5	C-5R	C-6	C-6RA	C6-RB	CF-3	C-7	CF-2	C1B	IGS39	IGS40	Avg slates
Mode	Layer	Layer	Layer	Layer	Layer	Layer	Layer/Sill	Layer	Layer	Layer/Sill	Layer/Sill	Host Fm
Type	Met	Met	Met	Met	Met	Met	Volc_F	Volc	Volc	Volc	Volc	n = 10 / 6
SiO <sub>2</sub>	33.9	50.6	43.5	58.1	40.5	54.8	74.5	42.9	43.8	39.43	44.89	52.93
TiO <sub>2</sub>	2.01	0.99	1.19	0.7	1.45	0.76	0.44	2.48	2.06	1.96	1.84	1.04
Al <sub>2</sub> O <sub>3</sub>	21.3	17.6	17	12.6	23.8	11.75	13.85	18.2	14.8	12.43	12.35	23.33
FeO	32.7	20.4	27.8	21	21.5	23.5	5.17	11.4	11.1	11.09	10.45	9.36
MnO	0.09	0.08	0.1	0.07	0.06	0.11	0.02	0.16	0.15	0.17	0.12	0.08
MgO	3.02	2.43	3.62	2.65	2.71	2.58	0.82	6.76	7.37	8.65	10.63	2.53
CaO	0.23	0.08	0.7	0.27	0.33	0.12	0.1	4.48	5.94	7.3	2.07	0.36
Na <sub>2</sub> O	0.2	0.36	0.04	0.05	0.53	0.03	0.47	5.33	4.43	3.72	4.08	1.22
K <sub>2</sub> O	0.63	1.43	0.16	0.26	2.62	0.05	2.47	0.19	0.04	0.05	0.07	3.74
P <sub>2</sub> O <sub>5</sub>	0.44	0.25	0.61	0.3	0.37	0.26	0.14	0.7	0.29	0.26	0.19	0.21
A/CNK	14.9	7.7	11.24	14.74	5.52	36.52	3.81	1.06	0.81	0.63	1.17	3.50
Ba	121.5	359	36.2	65.9	644	22.5	397	373	67.1	23	57	707.83
Cr	130	60	150	50	90	60	50	110	150	343.22	575.44	105
Cs	1.53	3.03	0.5	0.62	4.81	0.26	4.16	0.56	0.13	0.23	0.30	8.19
Ga	24.6	21.9	22.2	17.2	29.4	17.1	20.2	23.4	18.7	18.54	18.57	27.25
Hf	7.3	5.2	4.5	4.1	6.5	4.1	5	5.2	4.1	3.50	3.22	4.72
Nb	46.7	38.1	35.2	27.9	35.2	33.2	38.6	54.9	41.6	27.94	26.27	19
Rb	28.4	65.5	6.9	12.1	122	2.7	113.5	6.8	0.4	0.59	8.00	161.83
Sr	62.4	101	33.1	24.3	171	16.2	142	303	494	365	186	157
Ta	2.6	3.2	1.6	2.3	2.5	2.2	3.2	2.5	2.2	1.78	1.55	1.35
Th	15.5	13.25	11.05	10.5	19.4	11.9	15.35	5.27	3.81	2.77	3.30	20.73
U	3.08	3.84	2.4	2.99	3.92	3.06	4.59	1.31	0.88	0.75	0.75	3.22
V	146	100	139	90	173	94	43	248	204	150	162	132
Y	45.7	45.9	50.1	36.8	47.9	41.1	40.6	26.8	17.8	16.25	15.59	39.82
Zr	225	165	151	126	202	125	167	227	170	130	119	162.33
Cu	41	2	17	6	4	8	2	25	6	46.56	105.06	36.33
Li	290	200	280	200	200	220	70	190	150	233.33	190.74	138.33
Ni	37	30	39	22	34	32	12	49	73	152.31	250.58	56
Pb	13	9	25	16	29	24	15	2	3	2.35	3.42	27
Sc	18	13	17	11	15	13	9	20	20	24	27	18.67
La	35.2	27.2	30.1	20.4	51.9	22.9	41.1	38.9	29.1	23.07	22.73	63.27
Ce	71.2	58.7	61	43.4	101.5	51.6	78.6	74.2	52.4	47.79	46.85	122.58
Pr	9.47	7.41	8.39	5.52	12.15	6.76	9.43	8.96	6.02	5.48	5.48	14.37
Nd	39.1	29.4	35.2	21.9	47.2	28.2	34.5	36	23.5	23.59	21.39	54.12
Sm	9.3	7.45	9.19	5.47	10.3	6.87	7.55	7.5	4.73	4.95	4.56	10.76
Eu	2.03	1.16	2.05	1.23	1.87	1.46	1.16	2.28	1.38	1.58	1.37	2.01
Gd	10.25	7.07	10.05	5.99	10.35	6.54	7.9	6.18	4.75	4.42	4.17	8.84
Tb	1.78	1.35	1.8	1.05	1.72	1.24	1.37	1.06	0.68	0.63	0.56	1.32
Dy	10.65	8.55	10.1	7.06	10.3	8.41	7.72	5.1	3.97	3.36	3.11	7.48
Ho	2.22	1.81	2.17	1.37	1.99	1.79	1.65	1.04	0.8	0.61	0.61	1.45
Er	6.12	5.16	6.13	4.01	5.48	5.36	4.45	3.19	1.98	1.57	1.54	4.08
Tm	0.97	0.78	0.98	0.64	0.76	0.78	0.63	0.38	0.26	0.23	0.22	0.58
Yb	5.91	4.82	6.09	4.12	4.73	5.39	3.78	2.71	1.79	1.53	1.44	3.69
Lu	0.86	0.68	0.91	0.61	0.68	0.75	0.59	0.32	0.26	0.23	0.22	0.55

Major elements in wt% and trace elements + REE in ppm

*Volc*: volcanic rock, *Met*. metasomatic shale, *Volc F* felsic volcanic rock



**Fig. 5** Optical microscope images of the studied rocks. **a–d** The volcanic rocks (C-1B and CF-2) consist of plagioclase (Pl), chlorite (Chl) and variable amounts of carbonate phases. The carbonates grow over chlorite and plagioclase. Quartz + carbonate and carbonate vein-

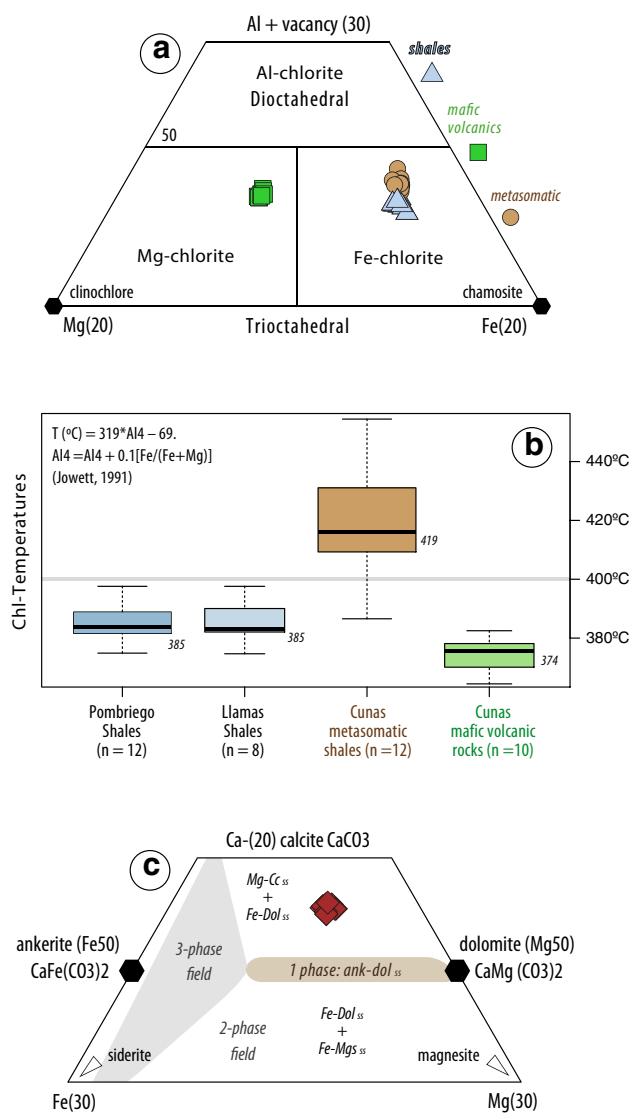
lets cut across the chlorite phases (**c, d**). Metasomatic shales (**e, f**) consist of chlorite + quartz ( $\approx 50:45\%$ ) and other minor components such as rock fragments (RF) and accessory phases like zircon (Zrn)

396 shales (2.77–2.83 a.f.u.). The calculated temperatures of  
 397 the metasomatic shales are above those calculated for the  
 398 rest of the rocks, having a mean of  $419 \pm 18$  °C and a total  
 399 range of 386–454 °C (Fig. 6b).

400 Carbonate minerals from the mafic volcanic rocks (M3  
 401 sample) consist of Fe-bearing dolomite or Fe–Mg calcite

(Table 2). Data from EMPA is projected in the triangular  
 402 diagram Fe–Ca–Mg where the fields of solid solutions and  
 403 carbonate phases are shown for  $T \approx 250$ –400 °C (Rosenberg  
 404 1967; Anovitz and Essene 1987) (Fig. 6c). The analyzed car-  
 405 bonates project in the two-phase field of coexistence between  
 406 calcite (Cc) or Mg-calcite solid solution phase (Mg-Cc<sub>ss</sub>) and  
 407





**Fig. 6** Chlorite and carbonate mineral chemistry. **a** Chlorite classification by Zane and Weiss (1998). **b** Chlorite temperature calculations by the method of Jowett (1991). Chlorite data from the host shales, sampled in other localities, was projected for comparison (unpublished data from Gómez-Fernández. See Table 1 for average values). **c** Ca-Fe-Mg diagram with the projection of the studied carbonates

408 Fe-dolomite solid solution phase (Fe-Dol<sub>ss</sub>). The analyzed  
 409 carbonates would not represent therefore a single phase but a  
 410 mixture of these two phases.

## 411 7 Geochemistry

412 The studied samples consist of volcanic-volcanoclastic  
 413 and metasomatic shales (Table 3). The volcanic and vol-  
 414 canoclastic rocks show SiO<sub>2</sub> contents of 39.43–44.89 wt.%

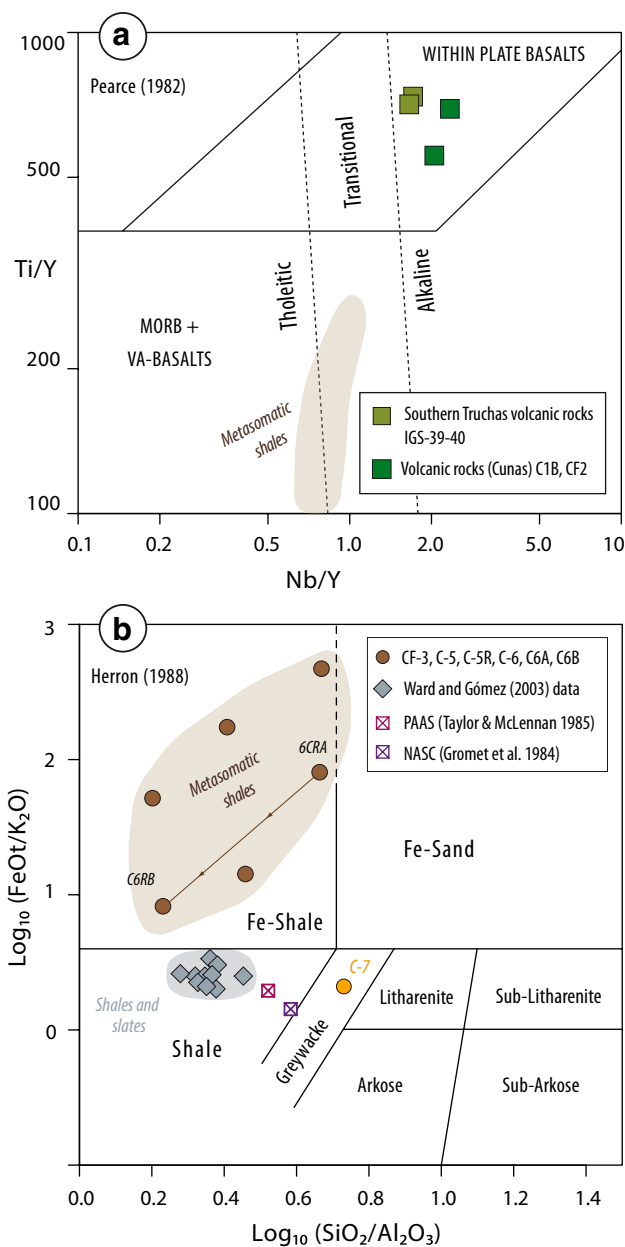
(average: 42.75 wt%) and Na<sub>2</sub>O + K<sub>2</sub>O of 3.7–5.5 wt%. 415  
 Their MgO (6.76–9.14 wt%), FeOt (10.45–11.40 wt%) 416  
 and CaO (2.07–7.30 wt%) range from moderate to elevated 417  
 values. Their TiO<sub>2</sub> contents are also high (1.84–2.48wt.%). 418  
 These features allow classifying these rocks as basalts, 419  
 trachybasalts and basanites. Two of them have slightly 420  
 anomalous A/CNK values (C-2 = 1.06, IGS-40 = 1.17), 421  
 probably due to alteration and partial Ca loss. Regarding 422  
 trace elements, these rocks have elevated Ti/Y (500–100), 423  
 Nb/Y (1.5–2.5), Th/Yb (1.5–2.5) and Nb/La (1–1.5) ratios, 424  
 indicative of their alkaline affinity. They plot in the field of 425  
 within-plate basalts (Fig. 7a) suggesting absence of sub- 426  
 duction/crustal contamination. 427

Within this volcanic-volcanoclastic group, sam- 428  
 ple C-7 makes a difference due to its felsic nature: 429  
 SiO<sub>2</sub> = 74.50 w.%, Na<sub>2</sub>O + K<sub>2</sub>O = 2.94 wt%, very low CaO 430  
 content (0.10 wt%), moderate to high MgO (0.82 wt%) and 431  
 very high peraluminosity (A/CNK = 3.81). This rock can 432  
 be classified as a rhyolite with a greywacke volcanoclastic 433  
 composition, following the criteria of Herron (1988) 434  
 (Fig. 7b). 435

The metasomatic shales have a wide SiO<sub>2</sub> range of 436  
 33.90–58.10 wt%, with the lowest SiO<sub>2</sub> values within 437  
 the sheared samples (Table 3). The most striking attrib- 438  
 ute of this group of rocks is the high FeOt content 439  
 (20.40–32.70 wt%; average: 24.48 wt%). The Al<sub>2</sub>O<sub>3</sub> con- 440  
 centration is heterogeneous but relatively high in some 441  
 cases (11.75–23.80 wt%, average: 17.34%). Due to the 442  
 very low CaO, Na<sub>2</sub>O and K<sub>2</sub>O abundances (< 1wt%), 443  
 the A/CNK values of these rocks are quite variable and 444  
 very high (5.52–36.52), being extremely peraluminous. 445  
 Positive correlations exist between FeOt and TiO<sub>2</sub>, Nb, 446  
 Sc and HREE. Due to its very high FeOt/K<sub>2</sub>O (log val- 447  
 ues: 0.91–2.67) and low SiO<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub> ratios (log val- 448  
 ues: 0.20–0.66), these rocks could be classified as Fe-Shales 449  
 (Fig. 7b) if a sedimentary origin is assumed. 450

The Luarca Formation, composed of shales-slates, 451  
 host the volcanic rocks and the metasomatic shales (Ward 452  
 and Gómez-Fernández 2003). These shales-slates have 453  
 relatively low contents of SiO<sub>2</sub> (49.55–55.27 wt%) and 454  
 CaO (0.18–0.72 wt%), and high Al<sub>2</sub>O<sub>3</sub> (23.17–26.08 wt%) 455  
 and FeOt (8.28–10.59 wt%) compared to standard shales 456  
 (Fig. 7b). Their A/CNK is high (3.14–3.97), thus being 457  
 very peraluminous rocks. The SiO<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub> ratio is low 458  
 (log values: 0.27–0.45) while the FeOt/K<sub>2</sub>O is slightly high 459  
 (log values: 0.30–0.52) (Fig. 7b). These geochemical 460  
 features allow us to classify these rocks as shales plotted 461  
 close to the limit with Fe-Shales and being more Fe-rich 462  
 compared to standards such as NASC (North American 463  
 Shale Composite, Gromet et al. 1984) and PAAS (Post 464  
 Archean Australian Shale, Taylor and McLennan 1985). 465

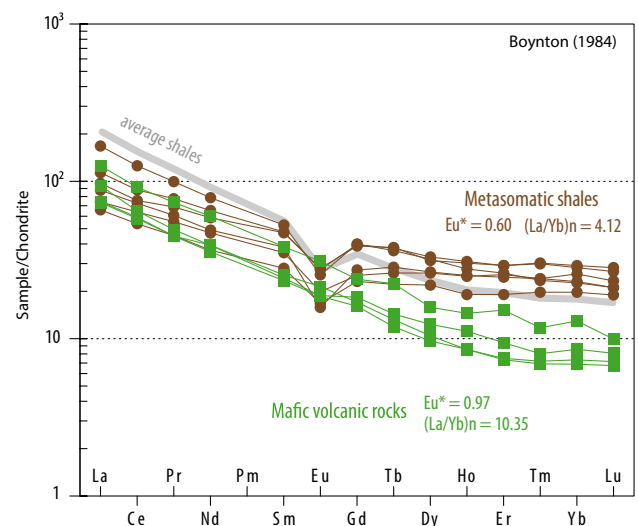




**Fig. 7** Classification diagrams used for the volcanic and metasomatic shales. **a** Tectonic setting diagrams based on the Ti/Y and Nb/Y ratios (Pearce 1982). **b** Classification diagram for sedimentary rocks (Herron 1988)

## 466 7.1 Rare earth elements

467 The contents and patterns of measured rare earth elements  
 468 (REE) show significant differences between the different  
 469 rocks studied (Fig. 8). The mafic volcanic rocks show a  
 470 fractionated pattern with La/Yb normalized values of 10.35  
 471 on average and very low to absent Eu negative anomalies  
 472 ( $Eu^* = [Eu_n / (Sm_n * Gd_n)^{0.5}] = 0.97$  on average). These data



**Fig. 8** Rare earth element normalized to chondrites (Boynton 1984) for the studied volcanic and metasomatic shales and average host shales (unpublished data from Gómez-Fernández–average in Table 3)

together with the heavy rare earths (HREE), below 10× 473  
 chondrite values, are typical features of alkaline basalts. 474

The REE patterns of the metasomatic shales are similar to 475  
 those displayed by the shales, with light rare earths (LREE) 476  
 fractionation and a flat HREE outline. The Eu anomaly of 477  
 these rocks is moderate ( $Eu^* = 0.60$ ) and very similar to that 478  
 shown by the host shales ( $Eu^* \text{Luarca Shales} = 0.63$ ). On the 479  
 contrary, the average fractionation of the REE is low ( $La_n /$  480  
 $Yb_n = 4.12$ ) and somewhat different to that of shales ( $La_n /$  481  
 $Yb_n \text{Luarca Shales} = 11.56$ ). 482

## 483 7.2 Comparison to bulk continental crust 484 composition

Spider plots normalized to the bulk continental crust of Tay- 485  
 lor and McLennan (1985) are presented in Fig. 9. These 486  
 show the differences and similarities of the trace element 487  
 contents between the volcanic, metasomatic and shale rocks. 488  
 The volcanic rocks show typical depletions in large ion litho- 489  
 philic (LIL) elements such as Rb, K, Ba and Cs, whereas Th and U 490  
 are similar to the bulk crust. High field strength (HFS) 491  
 elements (Nb, Ta, Sr, Zr, Hf, Ti) and LREE show values 492  
 above the bulk crust. On the other hand, the HREE are below 493  
 typical crustal values. The LIL elements of the metasomatic 494  
 shales display a pattern intermediate between volcanic and 495  
 shale compositions, with very variable K depletions and the 496  
 same Th and U values as the shales. Regarding the HFS ele- 497  
 ments, the metasomatic shales have similar Nb–Ta relations 498  
 to the volcanic rocks and closely similar Zr–Hf pattern as the 499  
 shales. Negative Ti anomalies and a strong Sr depletion are 500  
 common features with the shales. The REE pattern is more 501

**Fig. 9** Spider diagrams normalized to the bulk continental crust (Taylor and McLennan (1985). Shales data from Gómez-Fernández (unpublished average in Table 3) are compared to the studied metasomatic shales and volcanic rocks. Grey field is the total range of the data

502 similar to that of the shales but with slightly lower and vari-  
503 able LREE and higher HREE values.

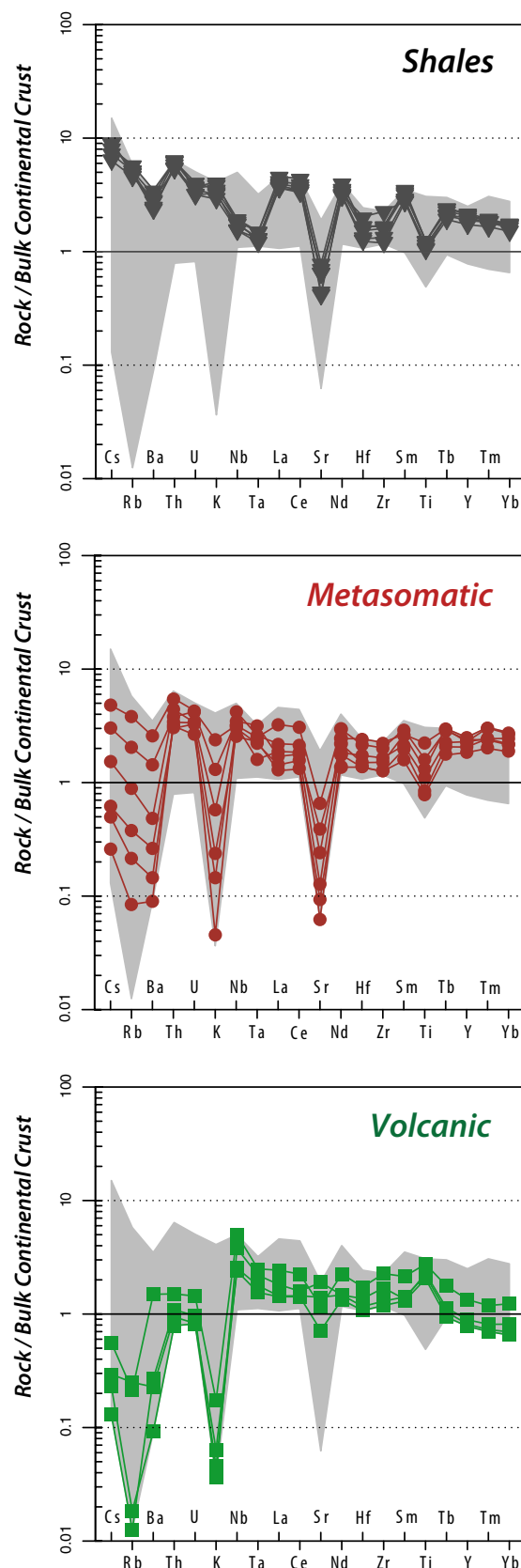
## 504 8 Discussion

### 505 8.1 Nature of the mantle source and geodynamic 506 context

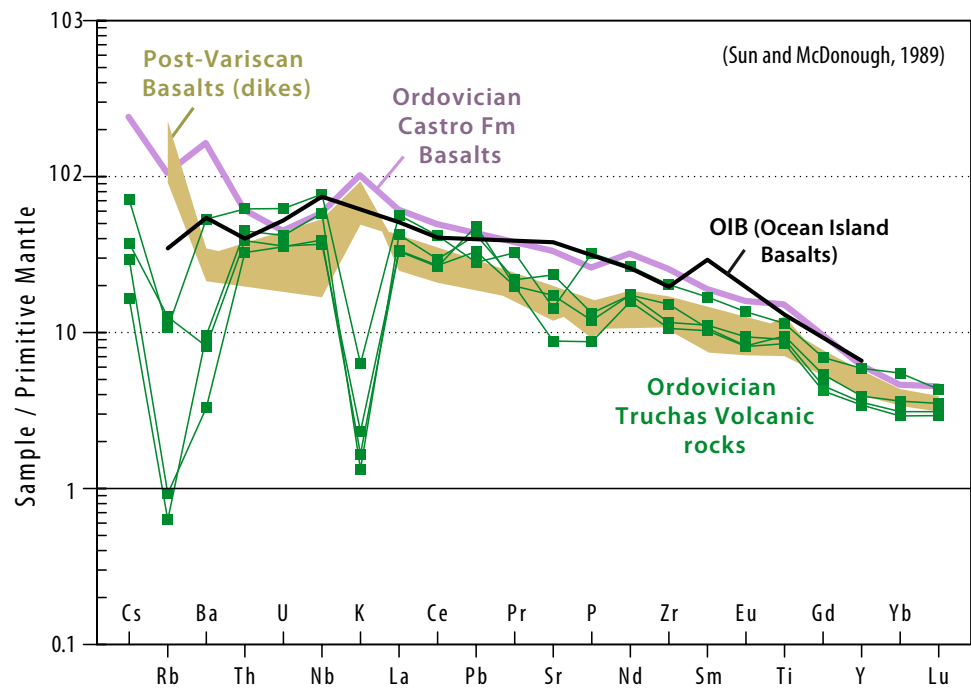
507 The studied Ordovician volcanic rocks from the Truchas  
508 Syncline have a geochemistry similar to OIB and within-  
509 plate basalts (Figs. 7a and 10). The main difference with  
510 OIB composition is the LILE depletion: Rb, K and variably  
511 Ba negative anomalies. No features of subduction or impor-  
512 tant crustal contaminations (Nb–Ta–Ti negative anomalies)  
513 are observed, except for some slight positive Pb anomalies  
514 (Fig. 10). Other Ordovician basalts and basalts of supposed  
515 post-Variscan age (Valverde-Vaquero et al. 2016; González  
516 Menéndez and Suárez, 2004) show a similar OIB-type pat-  
517 tern but with substantial differences in LILE compared to the  
518 studied Truchas basalts. The whole set of these basalts, even  
519 with their age differences, could have had a similar initial  
520 geochemistry affected by different processes: (i) More or less  
521 LILE enriched basalts whose content was later increased by  
522 hydrothermal fluids (González Menéndez and Suárez 2004)  
523 and (ii) subsurface/seafloor alteration and LILE leaching in  
524 the Truchas volcanic rocks. Other Iberian post-Variscan  
525 mafic dikes (Permian to upper Cretaceous) have a similar  
526 geochemistry except for the LILE depletions of the Truchas  
527 volcanic rocks (Gallastegui and Cuesta 2005; Orejana et al.  
528 2008). The source of the studied Truchas basalts was prob-  
529 ably a relatively enriched mantle unaffected by previous sub-  
530 duction processes. This protolith experienced low melting  
531 degrees. This is in agreement with the observed HFSE and  
532 REE fractionations and supports a tectonic regime of incipi-  
533 ent rifting and lithosphere thinning.

### 534 8.2 Carbonate phases in the volcanic rocks: 535 seafloor alteration vs. hydrothermalism— 536 metamorphism

537 Carbonates in some of these volcanic rocks are of interest  
538 because their origin could be related to H<sub>2</sub>O–CO<sub>2</sub> late-stage  
539 metamorphic fluids and these could act as transport agents  
540 of dissolved Au (Phillips and Evans 2004). Nevertheless,  
541 different hypotheses need to be considered in the first place.



**Fig. 10** Primitive mantle normalised trace element composition of the Truchas Syncline basalts. The shaded field corresponds to north Spain's post-Variscan basaltic dikes (González Menéndez and Suárez 2004). Ordovician Castro Fm. basalts are from Valverde-Vaquero et al. (2016). OIB average values are from Rollinson (1993) and references therein. Normalising values of primitive mantle are from Sun and McDonough (1989)



- 542 (i) The composition of the studied volcanic rocks is  
543 mainly mafic (basaltic) and their correspondent  
544 magmas could have considerable  $\text{CO}_2$  contents due  
545 to the high solubility of this phase in mafic magmas  
546 (King and White 2003). Nevertheless, volcanic rocks  
547 degas relatively fast, during and after emplacement,  
548 because of its low viscosity (Sparks 2003) imply-  
549 ing that  $\text{CO}_2$  deriving from the volcanic rocks is an  
550 unlikely source of  $\text{CO}_2$  in carbonates.
- 551 (ii) The studied volcanic rocks were emplaced into a shal-  
552 low marine sedimentary basin, where Ordovician  
553 shales were being deposited. In this paleo-environ-  
554 ment, carbonates could have formed by chemical  
555 precipitation from seawater and by alteration of Ca-  
556 bearing phases within the volcanic rock. Seafloor  
557 alteration/precipitation should have produced calcite  
558 as the main carbonate ( $\pm$  chlorite  $\pm$  albite) and  
559 this typically occurs on previous Ca-bearing miner-  
560 als, fracture fillings and rock-vesicles (Groves et al.  
561 1988). Furthermore, if this hypothesis is correct,  
562 most these volcanic samples should have carbon-  
563 ate phases because they have a fairly homogeneous  
564 basaltic composition and were expelled in the same  
565 marine environment and were exposed to the same  
566 seafloor alteration conditions. Of the different stud-  
567 ied samples, three have carbonate phases replacing  
568 primary plagioclase (Fig. 5).
- 569 (iii) Crustal fluids, produced by metamorphic dehydra-  
570 tion in deeper crustal locations, are composed of  
571  $\text{H}_2\text{O} + \text{CO}_2$  ( $\pm$  other minor salt components). These

572 fluids usually traverse upper crust formations where  
573 they cool and crystallize as quartz veins. Interaction  
574 with rocks that have Ca–Mg–Fe can lead to precipita-  
575 tion of different carbonates phases. The presence of  
576 Fe-dolomite, Mg–Fe calcite, ankerite and/or magne-  
577 site-siderite, besides calcite, should be indicative of  
578 metamorphic/hydrothermal processes (Groves et al.  
579 1988).

580 The petrography study shows that undeformed carbon-  
581 ates substitute chlorite and also appear in veins crosscut-  
582 ting the chlorite matrix (Fig. 5b–d). The observed carbon-  
583 ate compositions of Fe-bearing dolomite/Mg calcite favors  
584 a metamorphic origin for the fluids (Groves et al. 1988).  
585 These data thus support an origin of these carbonates as  
586 precipitates from crustal fluids generated after the chlorite  
587 ( $\approx 374 \pm 6^\circ\text{C}$ ) metamorphic growth. These fluids can flow  
588 through the crust in a pervasive way, or, more commonly,  
589 in more discrete zones where the fluids are collected. This  
590 could explain why some mafic volcanic rocks developed  
591 carbonate phases while others did not. Another possible  
592 explanation is that minor heterogeneities in the studied  
593 basalt compositions could have limited/augmented the  
594 availability of reactants and thus conditioned the devel-  
595 opment of carbonate phases.

### 8.3 T- $X(\text{CO}_2)$ modelling in the volcanic rocks

596 Phase diagram (pseudosection) modelling in the system  
597 NCaKFMASHC was done to investigate the formation of  
598



599 carbonate phases by crustal fluid ( $H_2O + CO_2$ )—rock interaction. The thermodynamic data base ds.55 of Holland and  
 600 Powell (1998) was used together with the following A–x  
 601 solution models: muscovite (Coggon and Holland 2002),  
 602 biotite (White et al. 2007), chlorite (Holland et al. 1998),  
 603 amphibole (Diener et al. 2007) and feldspar (Holland and  
 604 Powell, 2003). Carbonate phases calcite-dolomite-magnesite  
 605 (CcDo) were treated as a solid solution phase with the model  
 606 of Holland and Powell (2003). Ankerite and siderite were  
 607 considered as pure end members. The Pitzer and Sterner  
 608 (1994) equation of state was considered for modelling the  
 609  $H_2O$ – $CO_2$  (site mixing) behaviour.

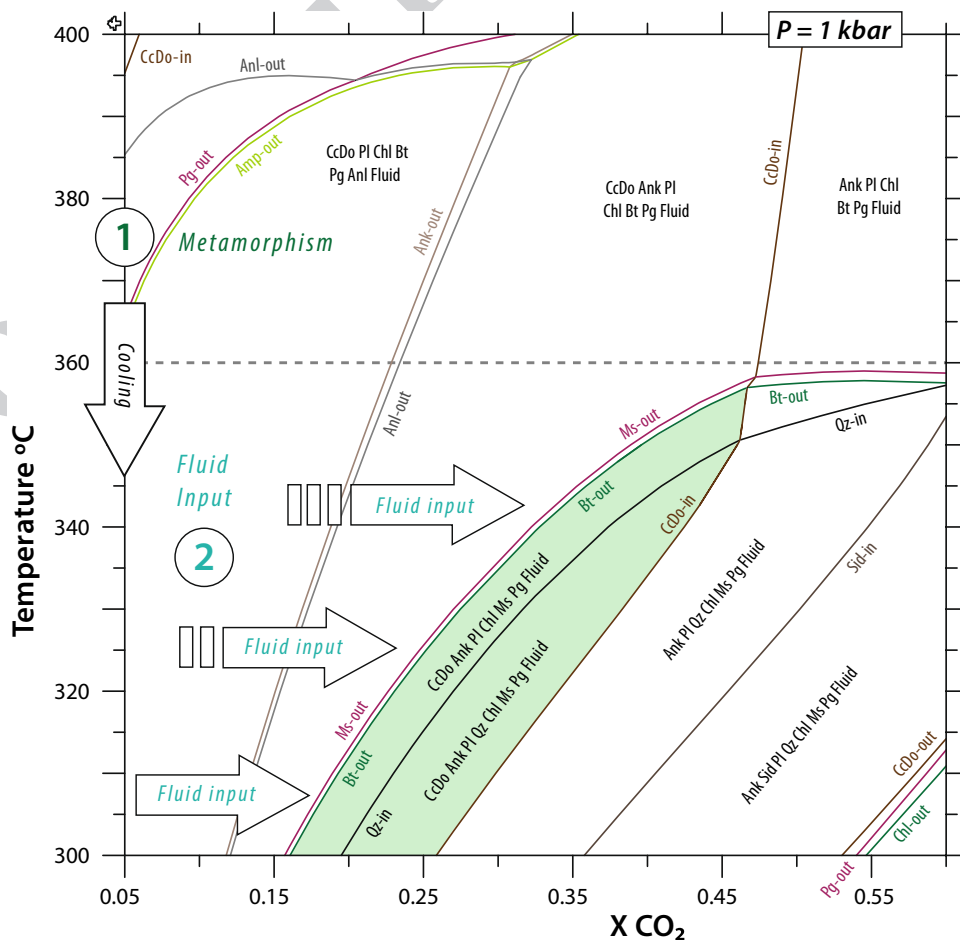
611 Theriak-Domino software (De Capitani and Petrakakis  
 612 2010), a Gibbs free energy (G) minimization code, was used  
 613 for calculations and diagrams. The input bulk rock-composition  
 614 is the sample CF2, representative of the volcanic rocks  
 615 with basaltic composition from the studied area (Table 3).

616 The results are presented in a T- $XCO_2$  equilibrium phase  
 617 diagram or pseudosection where temperature (T) is compared  
 618 to  $XCO_2 = nCO_2/nH_2O + nCO_2$ . The model was generated  
 619 with a fixed pressure of 1 kbar to model near surface  
 620 conditions (Fig. 11). Stable phase assemblages with the  
 621 lowest G values for the specific P–T–X conditions are

shown. High contents of  $H_2O$ – $CO_2$  were considered in order  
 to have fluid in excess. At the T range studied (300–400 °C)  
 the observed assemblage of albitic Pl + Chl + Fe-dolomite/  
 Mg-calcite  $\pm$  sericite  $\pm$  Qtz  $\pm$  K-feldspar is best matched by  
 the calculated assemblage of Pl(Ab) + Chl + CcDo + Ms-P  
 g  $\pm$  Qtz shown in the colored field of Fig. 11. The T- $XCO_2$   
 conditions for this field are varied because it has a positive  
 slope and therefore spans a considerable range of T  
 ( $\approx$  200–360 °C) and  $XCO_2$  (0.05–0.47). Higher T  
 values or lower  $XCO_2$  fluid composition will generate biotite  
 as a stable phase in the assemblage. Higher  $XCO_2$  contents  
 will lead to the increase of ankerite  $\pm$  siderite as the main  
 carbonate phases. The calculated modal amounts of the mineral  
 assemblage are close to those observed where Pl(Ab) + Chl  
 dominate (calculated Pl  $\approx$  42%, Chl  $\approx$  35%) followed by  
 the carbonates (calculated CcDo + Ank  $\approx$  11%) and sericite  
 phases (Ms + Pg  $\approx$  10%). Carbonates are slightly underesti-  
 mated while sericite phases are overestimated compared to  
 the observed modal amounts.

With this model we can only establish a maximum T  
 ( $\approx$  350–360 °C) below which the observed mineral assem-  
 blage was developed in a range of possible fluid  $XCO_2$   
 values. This temperature is slightly lower than the one recorded

**Fig. 11** T- $XCO_2$  Pseudosection at P=1 kbar. Mineral abbreviations—Pl plagioclase, Chl chlorite, CcD calcite-dolomite, Ank ankerite, Sid siderite, Ms muscovite, Pg paragonite, Anl analcime, Qtz quartz, Amp amphibole. Bulk rock in cation mol proportions is Si(71.39), Al(35.68), Mg(16.77), Fe(15.87), Ca(7.98), Na(17.19), K(0.40), C(1000-0),  $H_2O$ (0-1000). Stage 1 refers to the temperature estimated (Chl composition) for the low-grade metamorphic event. Stage 2 refers to the possible input of fluid  $H_2O$ – $CO_2$  input into the rock. See main text for further explanations



in the chlorite compositions ( $\approx 374$  °C) related to low-grade metamorphism (low greenschist facies). This, and the textural relations of carbonates, overprinting or cutting across the chlorite crystals, indicates that the growth of the carbonates took place at relatively lower temperatures and was an event of fluid ( $\text{H}_2\text{O}-\text{CO}_2$ ) infiltration that occurred after or during the waning stages of the low grade regional metamorphism.

Therefore, we propose a regional metamorphic event at stage 1 at  $T \approx 374 \pm 6$  °C (Fig. 11). The  $\text{XCO}_2$  fluid values were low ( $\text{XCO}_2 \leq 0.025$ ) and carbonate phases were either not stable or had very low modal abundances. This is supported by the fact that no chlorite-carbonate intergrowths or simultaneous chlorite-carbonate growths were observed in the samples. After this event and during cooling below  $\approx 350-360$  °C, an  $\text{H}_2\text{O}-\text{CO}_2$  fluid input occurred at stage 2 (Fig. 11). This caused the development of carbonates as crystals and veins cutting across the rest of the phases. The fluid composition could have been buffered by the breakdown of biotite to form muscovite/sericite. Although there are no records of this reaction in the rocks, if it had occurred, fluid chemistry would have changed upon cooling and exhumation of the rock and would have been buffered along the Bt-out reaction line in Fig. 11. Such behavior of changing fluid composition upon pervasive fluid infiltration is not uncommon in metamorphic rocks (Kleine et al. 2014).

Other scenarios where infiltration of  $\text{H}_2\text{O}-\text{CO}_2$  crustal fluids interacted with mafic rocks to produce carbonation and, in some cases, Au mineralization, have similarities and differences with the model here described.

A similar evolution was proposed by Kleine et al. (2015) and Kleine et al. (2016) in metabasaltic sills from Islay (Scotland) with  $\text{H}_2\text{O}-\text{CO}_2$  infiltration fluids being syn-metamorphic (but post peak) and carbonation only affecting the sill margins. On the other hand, a different sequence of events is proposed in the model of Elmer et al. (2007) where an Au-hydrothermal event and  $\text{H}_2\text{O}-\text{CO}_2$  fluid infiltration took place before the metamorphic peak conditions (440 °C).

In our study case it seems that the infiltration of  $\text{H}_2\text{O}-\text{CO}_2$  fluids and carbonation of the mafic rocks happened after the metamorphic peak (chlorite average temperature  $\approx 374 \pm 6$  °C) or during the waning stages of metamorphism. This is mainly based on the observation of carbonates minerals and quartz-carbonate veins cut across chlorite minerals (Fig. 5b–d).

$\text{H}_2\text{O}-\text{CO}_2$  fluids, with the suitable ligands ( $\text{HS}^-$ ,  $\text{Cl}^-$ ), are known to be carriers of dissolved gold as bisulphide ( $\text{Au}(\text{HS})_2^-$ ) or chloride complexes ( $\text{AuCl}_2^-$ ). The observed Au mineralization in this region (Truchas Syncline) could be related to this  $\text{H}_2\text{O}-\text{CO}_2$  input. The relatively moderate temperatures estimated for this fluid (< 350–360 °C) are similar to those described in Au mineralization fluids from other

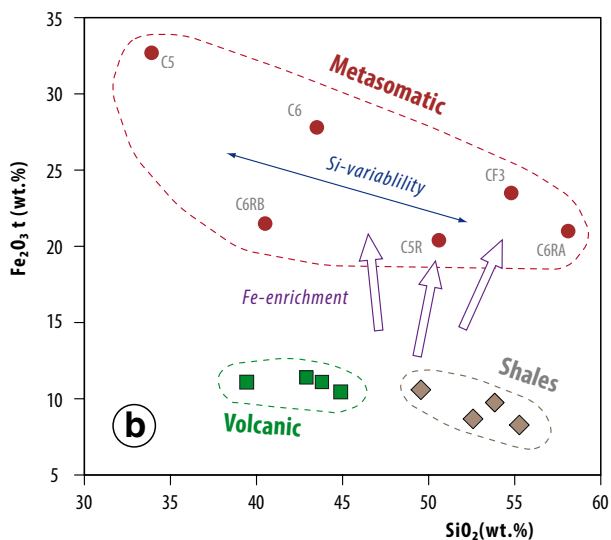
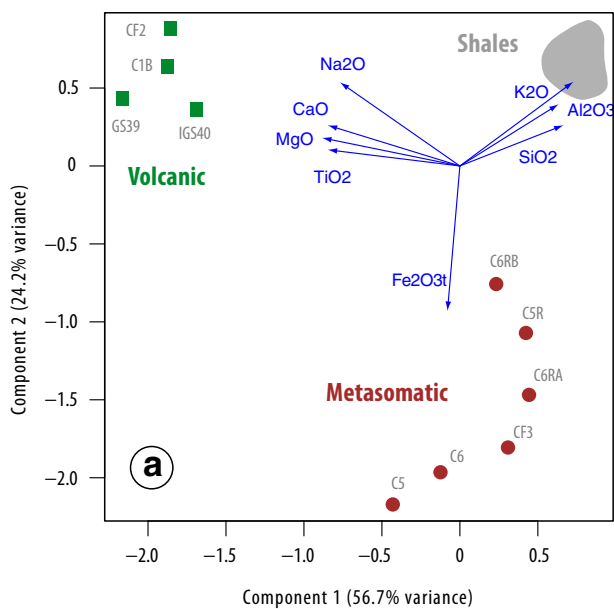
regions ( $\geq 200-300$  °C; Phillips and Evans 2004; Phillips and Powell 2010).

#### 8.4 Origin of the metasomatic shales

Previously, the lithological layers that we classified as metasomatic shales in this study were reported to be of volcanic origin (Suárez et al. 1994). However, here we show evidence that these rocks lack chilled margins, scoriaceous upper layers or observable igneous textures such as phenocrystals (Fig. 4). In thin section, the mineral assemblage observed is chlorite + quartz  $\pm$  rock fragments and accessory zircon (Fig. 5e,f) which casts doubts about a sole volcanic origin. The chlorite chemistry (Fig. 6) and whole rock geochemical data indicate an origin closer to a shale  $\pm$  sandstone composition with a minor volcanic component (Fig. 9). The rare composition of these rocks indicates that they are neither volcanic (igneous) nor completely sedimentary and instead, should be considered as metasomatic. The more deformed samples have low  $\text{FeOt}/\text{K}_2\text{O}$  and  $\text{SiO}_2/\text{Al}_2\text{O}_3$  ratios and the deformation effect on the geochemistry is shown in Fig. 7b (deformation trend).

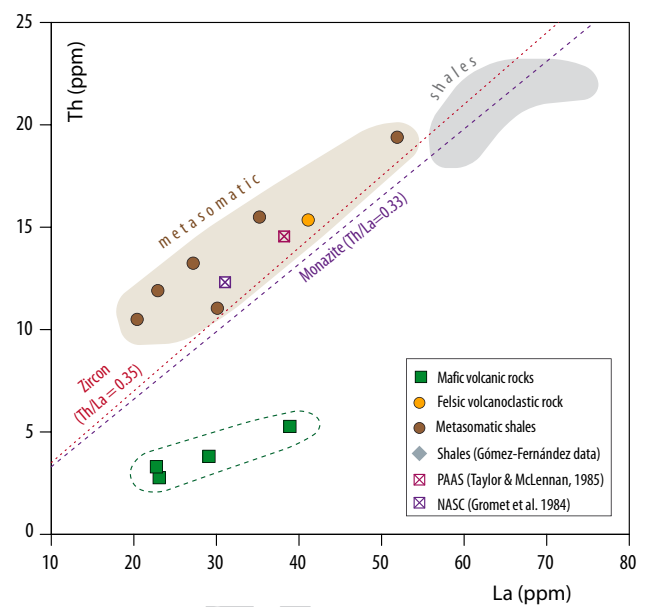
To further explore its geochemistry, we performed principal component analysis (PCA) including the whole set of studied rocks: metasomatic, volcanic and shale compositions (shales from Ward and Gómez-Fernández 2003 and unpublished data from Gómez-Fernández). PCA shows the geochemical variance grouped in a small set of components. The first two components account for 75% of the variance (Fig. 12a). The first one is dominated by the variables that separate the geochemistry of volcanic rocks from that of the shales:  $\text{SiO}_2$ ,  $\text{Al}_2\text{O}_3$ , and  $\text{K}_2\text{O}$  typical of the shale compositions and  $\text{CaO}$ ,  $\text{MgO}$ ,  $\text{TiO}_2$  and  $\text{Na}_2\text{O}$  that correspond to the mafic rocks. Metasomatic shales are closer in this 8-dimensional space to the shales. The second component is dominated by  $\text{Fe}_2\text{O}_3$  negative loading with minor contributions of  $\text{K}_2\text{O}$ ,  $\text{Na}_2\text{O} \pm \text{SiO}_2 \pm \text{Al}_2\text{O}_3$ . This shows the importance of the iron enrichment in these rocks accompanied with a decrease in alkalis ( $\text{Na}_2\text{O}$ ,  $\text{K}_2\text{O}$ ). This alkalis decrease, without an accompanying  $\text{Al}_2\text{O}_3$  increase, is responsible for the extreme peraluminosity of these rocks. Within the metasomatic shales, there is also a high  $\text{SiO}_2$  variability (Fig. 12b).

A high affinity of the metasomatic shales and shale compositions can also be seen with the Th and La diagram (Fig. 13). The Th/La ratios of these rocks are closer to those of zircon (0.35) and monazite (0.32) and different from those defined by the volcanic whole-rock data. We have also performed isocon analysis (Grant 1986; 2005) considering both a shale source (average composition in Table 3) and a volcanic protolith (CF-2) to generate an altered metasomatic rock (CF-3). The 1:1 isocon black line and the isocon dashed lines obtained considering  $\text{TiO}_2$ , Zr and MnO as immobile elements were used to define the region of



**Fig. 12** **a** Results of Principal Component Analysis (PCA) of the whole data set (variables: major elements). In the bi plot, vectors express the correlations of the variables to the principal components: variables that are close to the center of the plot are less important (lower loading) for the first components. The Individual samples contribution to variance is projected as well. A sample that is on the same side of a given variable (vector) has a high value for this variable. **b**  $\text{SiO}_2$  vs.  $\text{Fe}_2\text{O}_3\text{t}$  showing both the Fe- enrichment and Si variability of the metasomatic rocks compared to both the shales and volcanic rocks

749 least element mobility (Fig. 14). The results show that in  
750 all cases the metasomatic rocks gained  $\text{Fe}_2\text{O}_3\text{t} \pm \text{Nb}$  (shale  
751 precursor) or  $\text{Fe}_2\text{O}_3\text{t} + \text{Pb} \pm \text{SiO}_2$  (volcanic precursor). On  
752 the other hand, there are net losses in:  $\text{K}_2\text{O}$ ,  $\text{Na}_2\text{O}$ ,  $\text{CaO}$ ,  $\text{Rb}$ ,  
753  $\text{Sr}$ ,  $\text{Ba}$ , and  $\text{LREE}$  (shale precursor, Figs. 14a, b) or  $\text{CaO}$ ,  
754  $\text{Na}_2\text{O}$ ,  $\text{MgO}$ ,  $\text{Sr}$ ,  $\text{Ba}$ ,  $\text{V}$  (volcanic precursor, Fig. 14c, d).

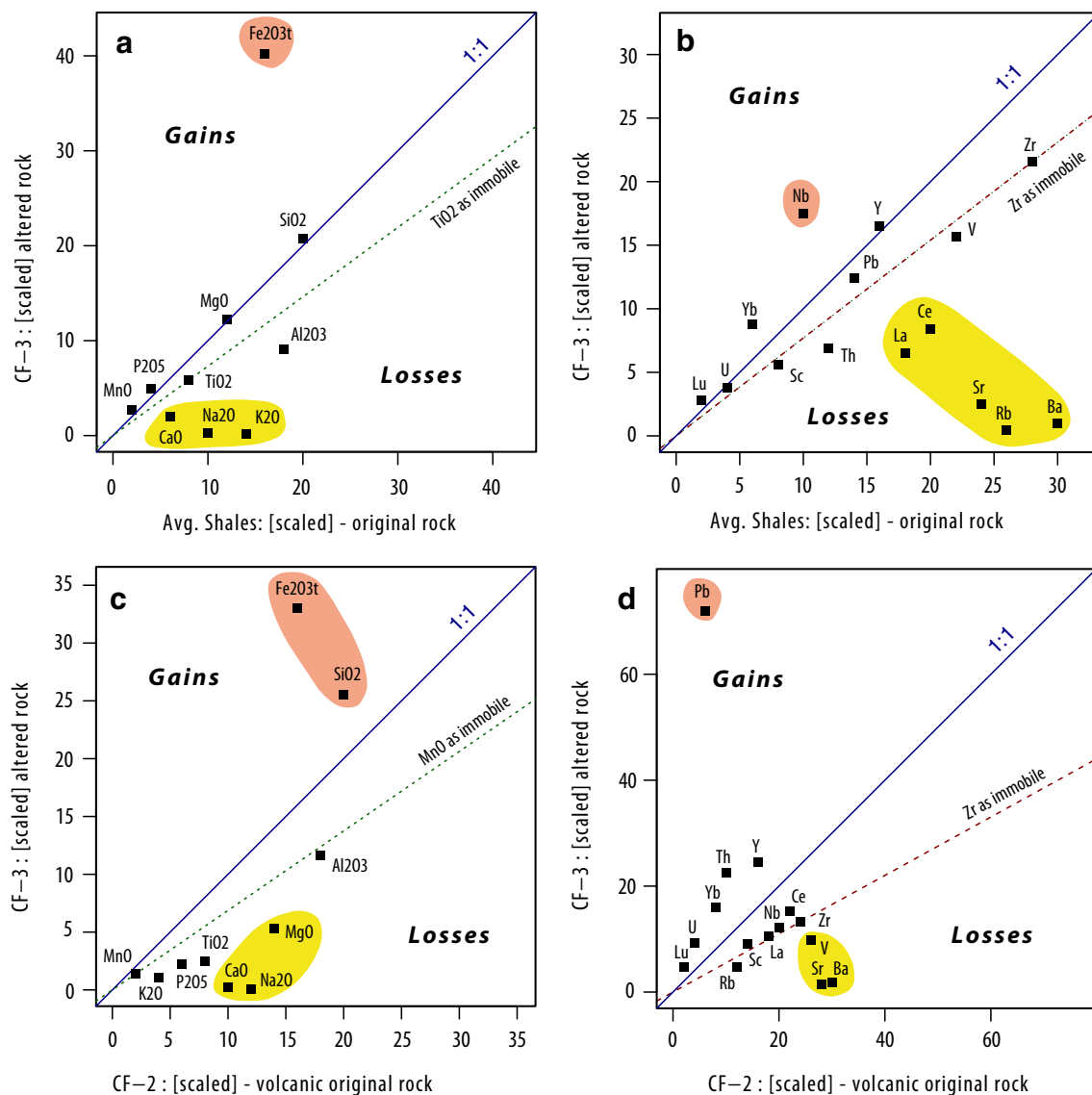


**Fig. 13** Th vs. La diagram showing the whole rock variability of volcanic, metasomatic, host shales and standard shales. Zircon and monazite Th/La ratios are projected as dashed lines for comparison

The common ground is the marked Fe enrichment and the  $\text{Na}_2\text{O} + \text{CaO} + \text{Sr} + \text{Ba}$  depletions.

Some of the borders of the metasomatic shales appear strongly foliated in comparison with its interior. We sampled both parts to investigate if there was any geochemical difference (from the border to the interior: C-6RB, C-6, C-6RA; Fig. 15). Since these rocks have a volcanic component, primary mineral-geochemical relations need to be considered. Mineral redistribution during a volcanoclastic emplacement could involve enrichments in Al, Si, Na, Ba, Rb and Sr in the body interior and higher concentrations of Ti, Fe, Mg, Mn, HFS elements and REE towards the margins (Kleine et al. 2016). In the study profile we have normalized the margin rocks (foliated) to the interior sample (Fig. 15). The most marginal and foliated sample (C-6RB) shows depletions in  $\text{SiO}_2 \pm \text{MnO}$  and enrichments in most of other elements, especially in  $\text{TiO}_2$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{CaO}$ ,  $\text{Na}_2\text{O}$ ,  $\text{K}_2\text{O}$ ,  $\text{Rb}$ ,  $\text{Sr}$ ,  $\text{Ba}$  and  $\text{LREE}$ . Overall, this pattern is not consistent with a magmatic mineral-element redistribution. The geochemistry variation from the rock core towards the edges does not have a specific trend because the intermediate sample (C-6) does not have an intermediate geochemistry. The strong foliation observed suggests that the shearing associated with Variscan tectonics might have modified the chemistry of the rock margins by dissolving  $\text{SiO}_2$  rich minerals (mainly quartz  $\pm$  feldspars) with pressure solution and leached  $\text{SiO}_2$  locally from the most deformed rocks. The rest of the elements would have had less mobility in these conditions and therefore increased its concentration by a corresponding residual enrichment.





**Fig. 14** Isocon diagrams (Grant 1986; 2005) of the average shales (a, b) and volcanic rock CF-2 (c, d) as possible protoliths (original rocks) for the metasomatic shales (CF-3 altered rock). The solid line

is the 1:1 reference line. Dashed lines are the slopes based on possible immobile elements (Ti, Zr, MnO)

## 785 8.5 Sequence of geological processes

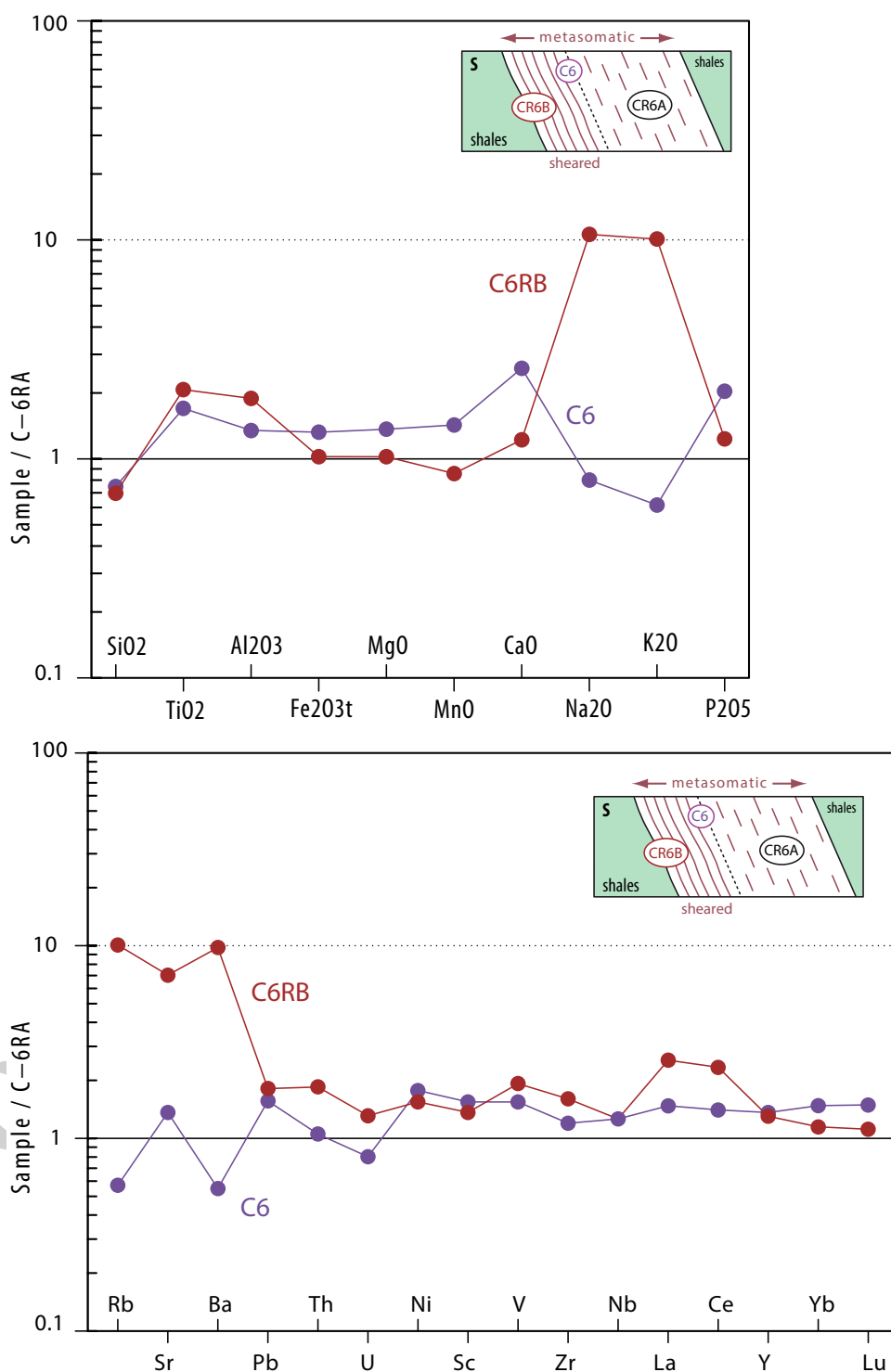
786 The data here presented, from field observations, petrography,  
787 geochemistry and T-XCO<sub>2</sub> modelling, suggest that these  
788 Ordovician volcanic rocks were emplaced in a relatively  
789 shallow marine basin as lavas and volcanoclastic deposits.  
790 Very rapid cooling generated the scoriaceous layers char-  
791 acteristic of some of these rocks. The magma composition  
792 could have been mafic with minor dacitic-rhyolitic pulses.  
793 The rapid cooling of these bodies, due to their relatively  
794 small thickness, precluded a metamorphic aureole develop-  
795 ment in the country rock shales. Preservation of orthid bra-  
796 chiopods in some volcanoclastic layer reinforces a Middle

Ordovician age and a shallow marine setting for the volcanic  
797 deposits.

Possible seafloor alteration is mostly obliterated by the  
798 later Variscan metamorphism. It probably consisted in  
799 some chlorite ± albite development. This process could have  
800 affected all these volcanic rocks in a similar way and with  
801 similar intensity.  
802  
803

The metasomatic shales, previously mapped as volcanic  
804 layers, probably represent a mixture of shales ± sandstones  
805 and minor volcanic components. The significant SiO<sub>2</sub> deple-  
806 tion in the foliated rock margins could have happened by  
807 mineral dissolution (quartz ± feldspars) produced by the later  
808 Variscan deformation (shearing).  
809

**Fig. 15** Major and selected trace elements composition of the foliated metasomatic shales (C-6, C-6RB) normalized to the less deformed, non-foliated sample (C-6RA). Insets show the relative position of the samples in the outcrop



810 The Variscan Orogeny folded and metamorphosed these  
 811 rocks up to greenschist grade as recorded in the Chl tem-  
 812 perature estimations ( $\approx 374 \pm 6$  °C). We propose that, dur-  
 813 ing this orogenic event, after the metamorphic peak, crustal  
 814 H<sub>2</sub>O–CO<sub>2</sub> fluids flowed through some of these volcanic  
 815 rocks and produced fluid–rock interactions with Ca–Fe–Mg  
 816 minerals at hydrothermal temperatures (< 350–360 °C).

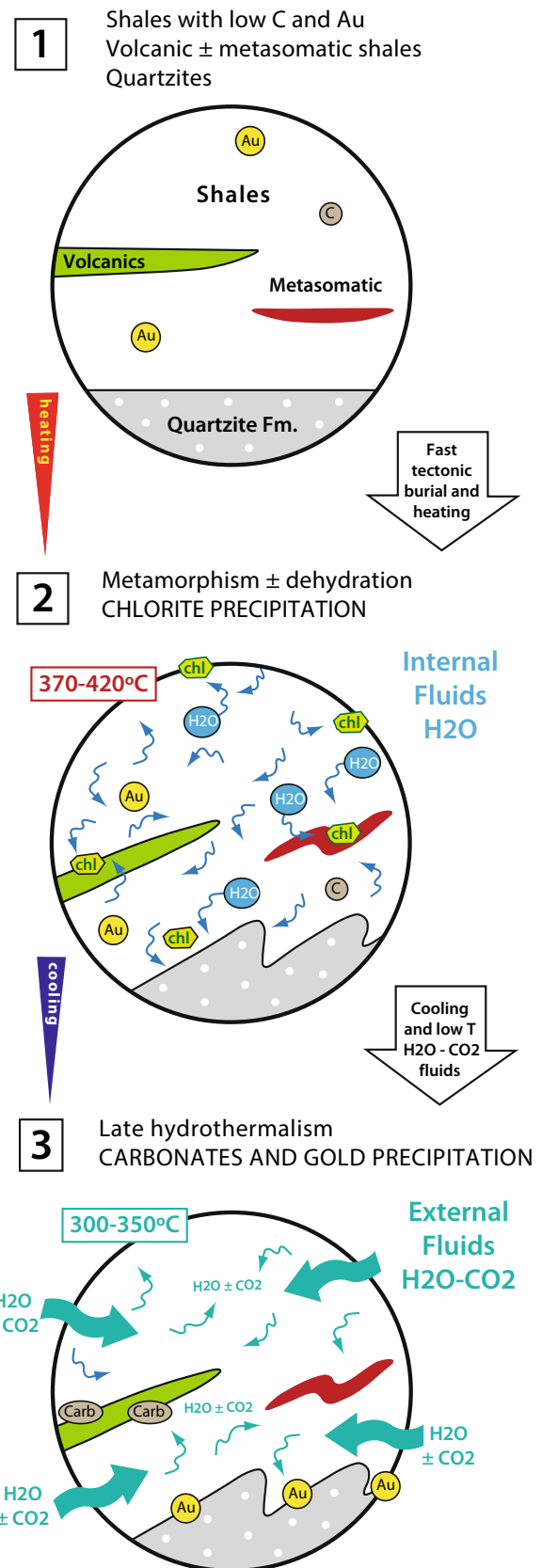
Depending on the XCO<sub>2</sub> of the fluid, different carbonate  
 phases could have been produced (as shown in the modelling  
 of Fig. 11). Gold could have been leached and transported by  
 some of these fluids, giving rise to economic gold precipita-  
 tion in some specific locations. The occurrence of carbonate  
 phases in some of the country rocks (sedimentary/volcanic;  
 Gómez-Fernández et al. 2009) could indicate the passage of

817  
 818  
 819  
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 822  
 823

**Fig. 16** Conceptual model in three stages for the Variscan metamorphism and late hydrothermalism that led to the carbonate precipitation and possible gold mineralization in the studied zone. See the main text for explanations

824 crustal fluid with mixtures of  $H_2O-CO_2$  that can also carry  
825 dissolved Au to produce mineralization.

826 Based on the data presented, a conceptual model has been  
827 described showing the possible evolution of these rocks in  
828 relation to the Variscan metamorphism, carbonate precipita-  
829 tion and gold mineralization (Fig. 16). In a first stage, the  
830 original upper crustal rocks of this zone were composed of  
831 volcanic + volcanoclastic layers ± metasomatic layers and  
832 surrounding shales (Fig. 16). The shales have low initial  
833 amounts of organic C and dispersed low amounts of Au  
834 among diagenetic pyrite (Gómez-Fernández et al. 2019;  
835 Cunningham et al. in prep). The Variscan orogeny produced  
836 heat leading to T increases up to  $\approx 374-420$  °C. This caused  
837 a widespread chlorite precipitation from dehydration reac-  
838 tions of previous phyllosilicate minerals within the shales  
839 and metasomatic rocks. Interactions of these  $H_2O$ -rich fluids  
840 with the volcanic and metasomatic rocks produced a retro-  
841 grade metamorphism of mafic minerals of the volcanic rocks  
842 and in the volcanic components of the metasomatic rocks  
843 (Fig. 16, stage 2). The low C content of the shales probably  
844 precluded the generation of  $H_2O+CO_2$  fluids at this stage, as  
845 indicated by the lack of carbonate phases growing simultane-  
846 ously with chlorite. After the metamorphic peak, at the ini-  
847 tial cooling stages ( $T \leq 350-360$  °C) infiltration of external  
848 and probably deeper  $H_2O+CO_2$  fluids occurred. These flu-  
849 ids interacted with the different rock compositions (Fig. 16,  
850 stage 3). In the case of the volcanic rocks, having significant  
851 CaO contents (in plagioclase and mafic primary and second-  
852 ary minerals), the reaction produced carbonate phases (Mg-  
853 calcite/Fe-Dolomite). The interaction of these fluids with  
854 the shales and metasomatic rocks did not cause carbonate  
855 precipitation because of the very low CaO contents of these  
856 rocks (0.08–0.36 wt%). For precipitation of pure Mg-Fe-  
857 carbonate phases (magnesite, siderite), higher  $XCO_2$  (mole  
858 fractions) would be needed in the fluid. The gold mineraliza-  
859 tions known around this area (Gómez-Fernández et al. 2012)  
860 could be related to these late  $H_2O+CO_2$  fluids that trans-  
861 ported dissolved Au (as bisulphide complexes  $Au(OH)_2^-$ ).  
862 The ultimate origin of the gold is unknown, it could have  
863 been leached in part from the surrounding shales, and/or  
864 being leached from deeper geological settings where these  
865 late crustal fluids came from.





866 **9 Conclusions**

867 The Ordovician volcanic and volcanoclastic rocks of NW  
868 Spain are well represented in the Truchas Syncline and in  
869 the studied area of this work (SE Truchas Syncline, Cunas  
870 village). Here, these rocks are hosted and interbedded within  
871 Ordovician shales (Luarca shale Fm.) and mainly consist  
872 of within-plate alkaline basalts extruded in a marine pas-  
873 sive margin. Such features are consistent with an extended  
874 passive continental margin linked to continental rifting  
875 processes.

876 Fossil occurrences (orthid brachiopods) within some of  
877 the volcanoclastic layers corroborate the Middle Ordovi-  
878 cian age for these rocks and its shallow marine conditions  
879 of emplacement.

880 Metasomatic shales occur as distinctive layers within  
881 the shales succession. These chlorite + quartz rocks show  
882 important iron enrichments and extreme peraluminosity. Its  
883 protolith could be a shale ± sandstone with minor volcanic  
884 components.

885 Variscan greenschist metamorphism affected these rocks  
886 promoting the widespread chlorite growth at  $T \approx 374 \pm 6$  °C,  
887 from primary mafic phases not currently preserved. Simul-  
888 taneous deformation processes are observed in some cases  
889 where the chlorite has a slight preferred orientation.

890 Subsequent to this metamorphism and deformation, or in  
891 its waning stages, infiltration of H<sub>2</sub>O–CO<sub>2</sub> fluids occurred  
892 through veins of variable scale and nature (Qz + Carbon-  
893 ate- and Carbonates-veins). These crustal fluids interacted  
894 with the volcanic rocks and particularly with its Ca–Fe–Mg  
895 phases, to produce carbonate minerals: Fe-bearing dolo-  
896 mite—Mg-bearing calcite at  $T \leq 350$ – $360$  °C. It is proposed  
897 that these H<sub>2</sub>O–CO<sub>2</sub> fluids could have been responsible for  
898 the transportation and deposition of gold mineralizations  
899 located nearby.

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